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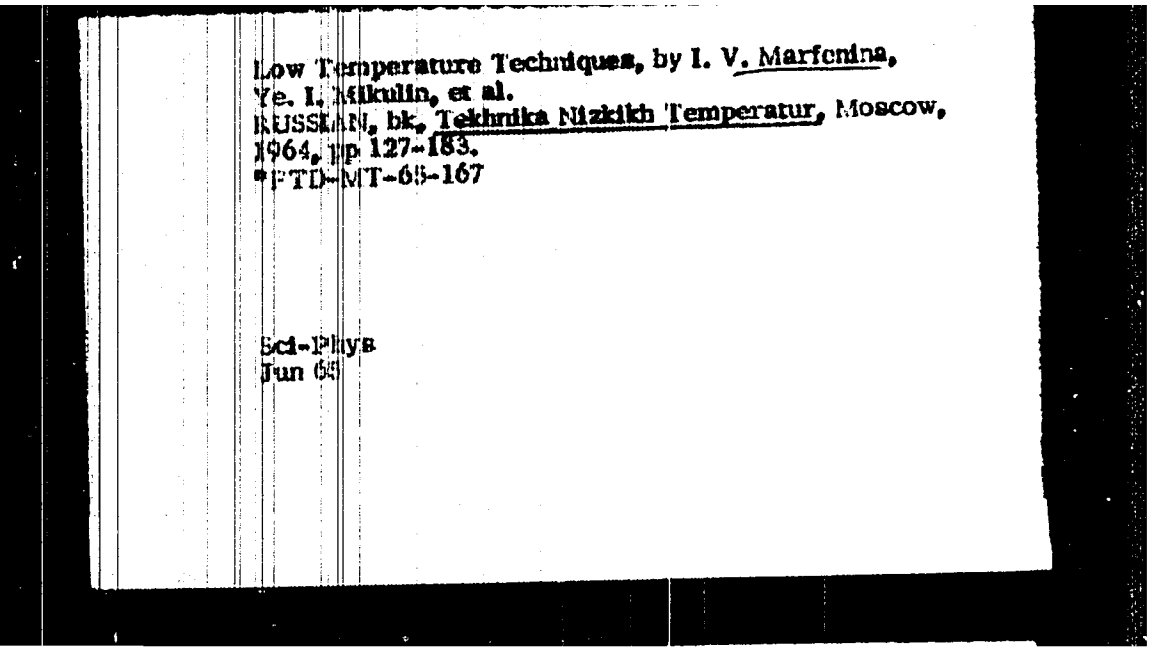

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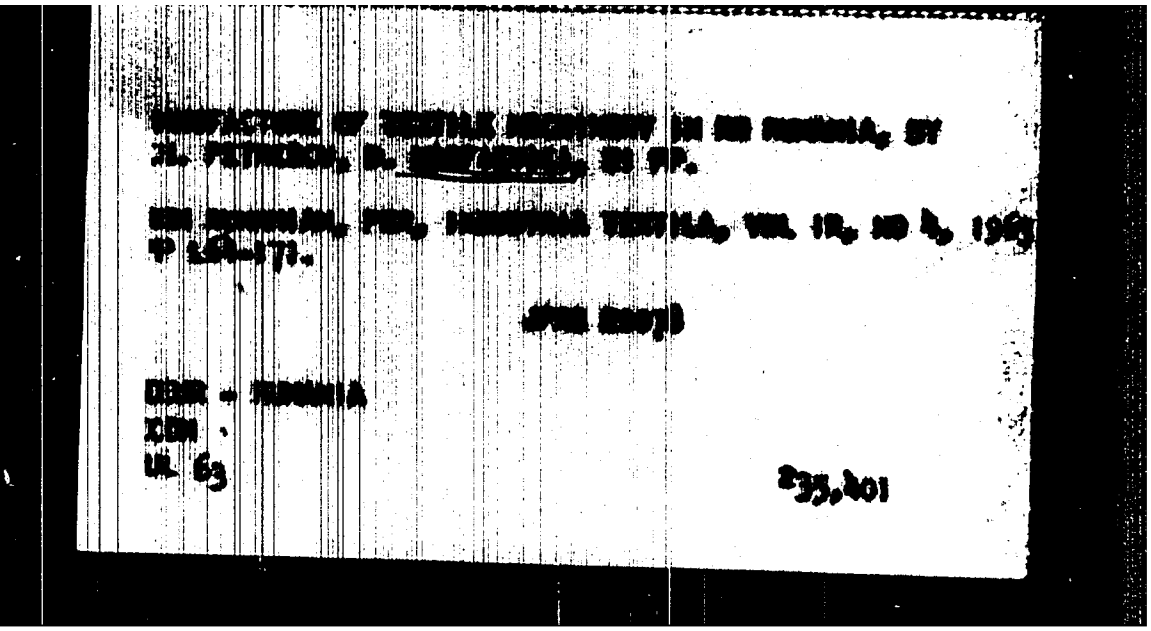

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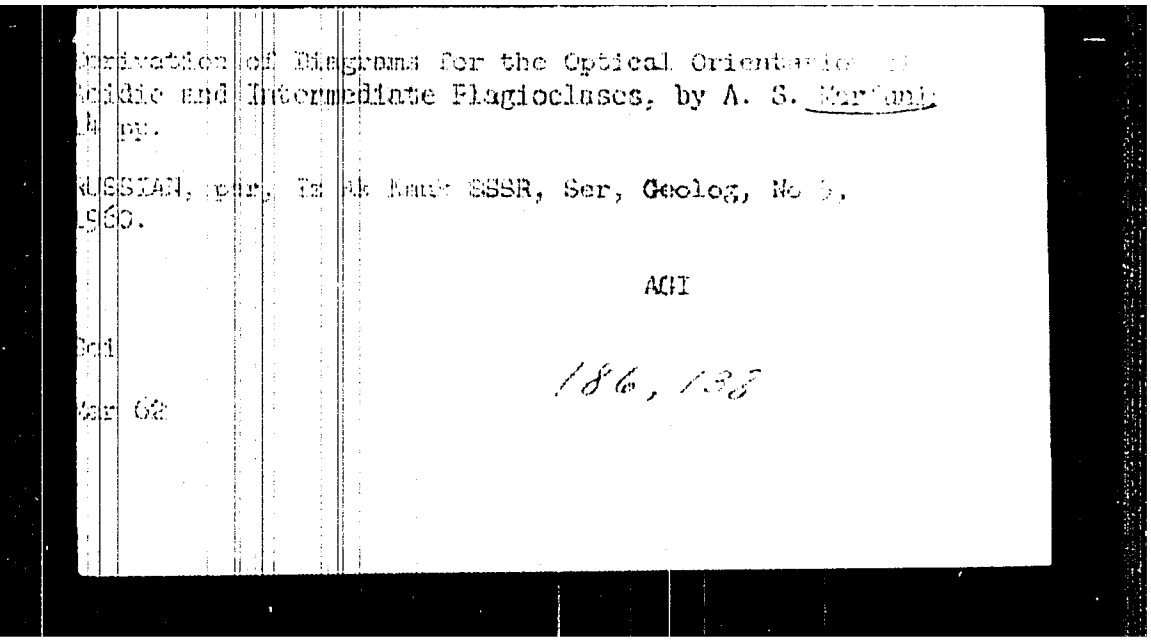

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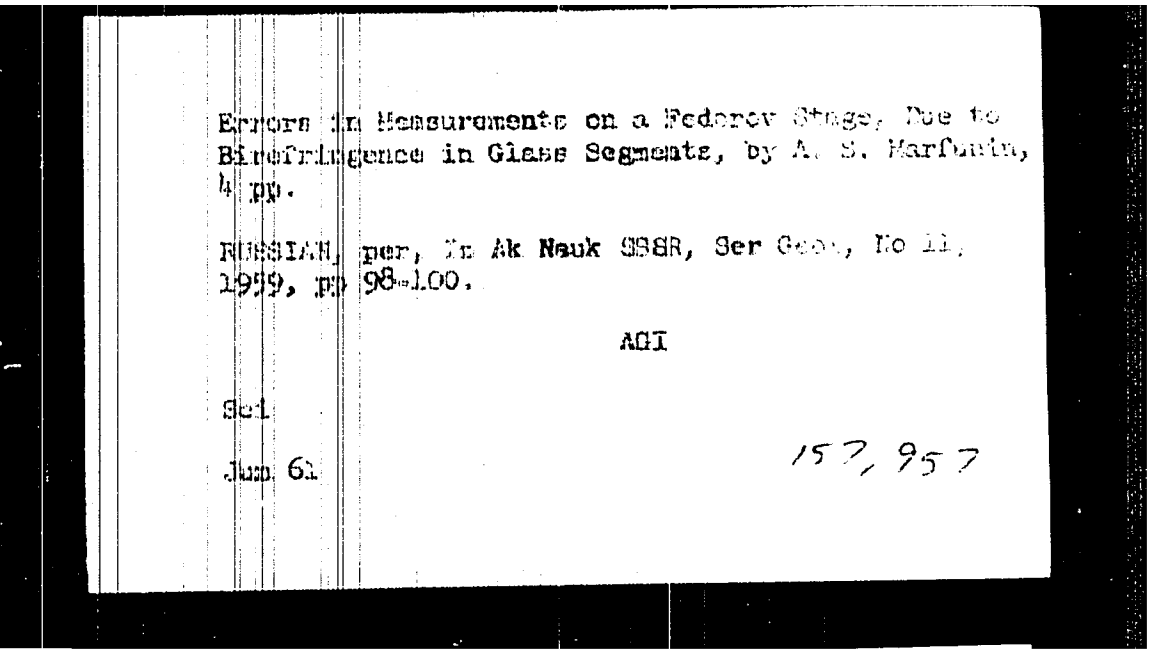

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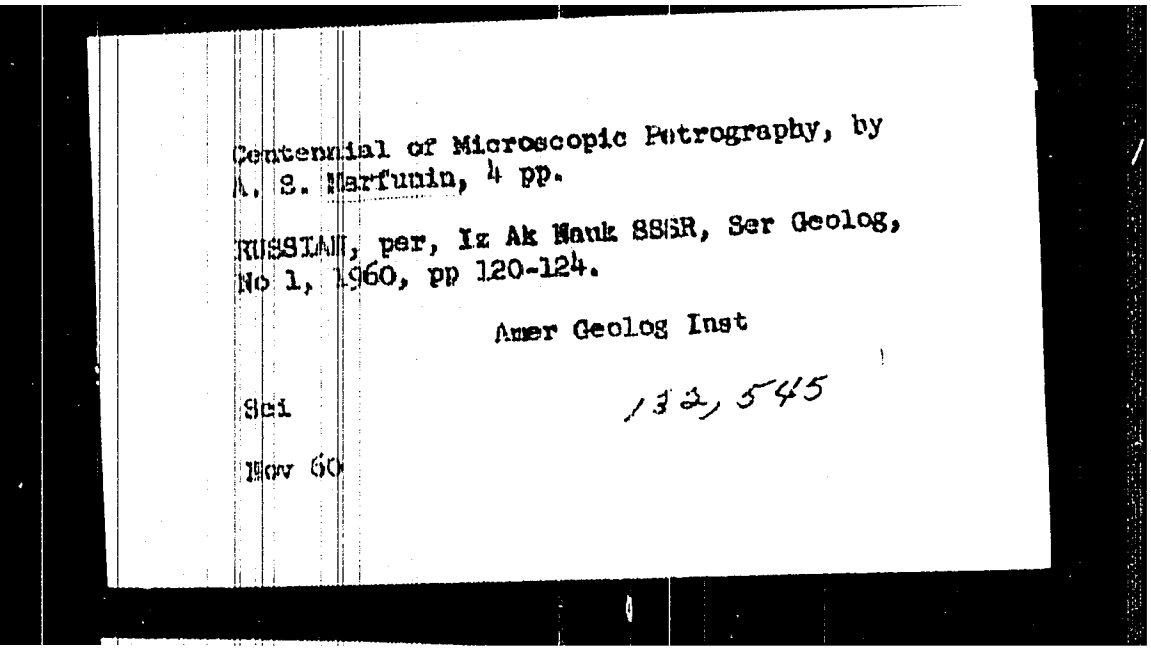

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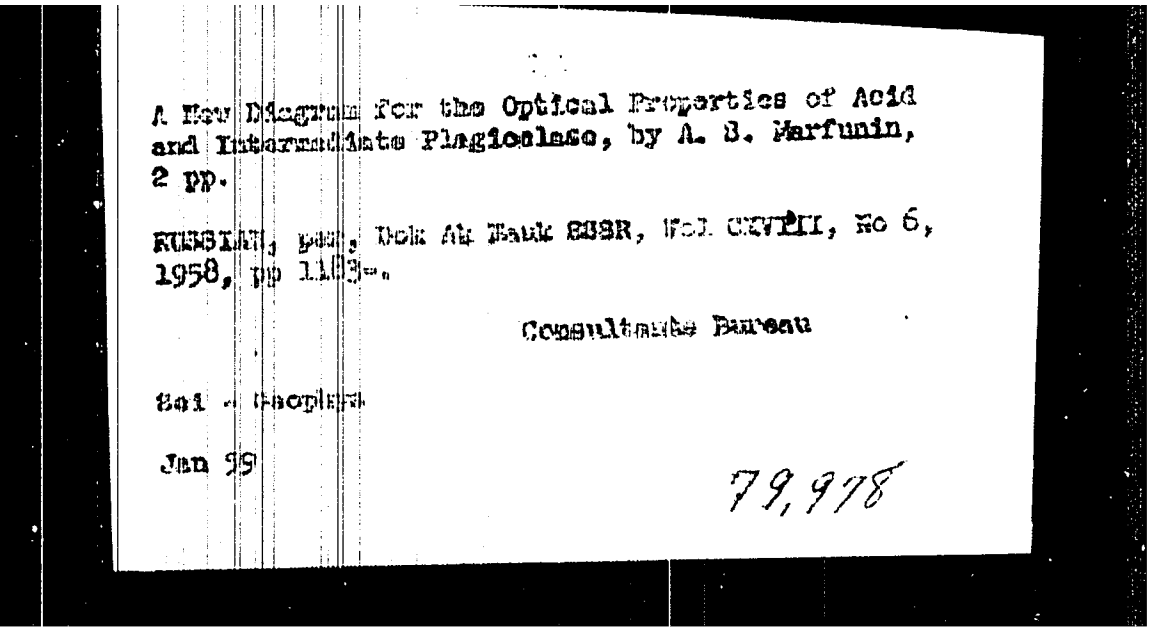

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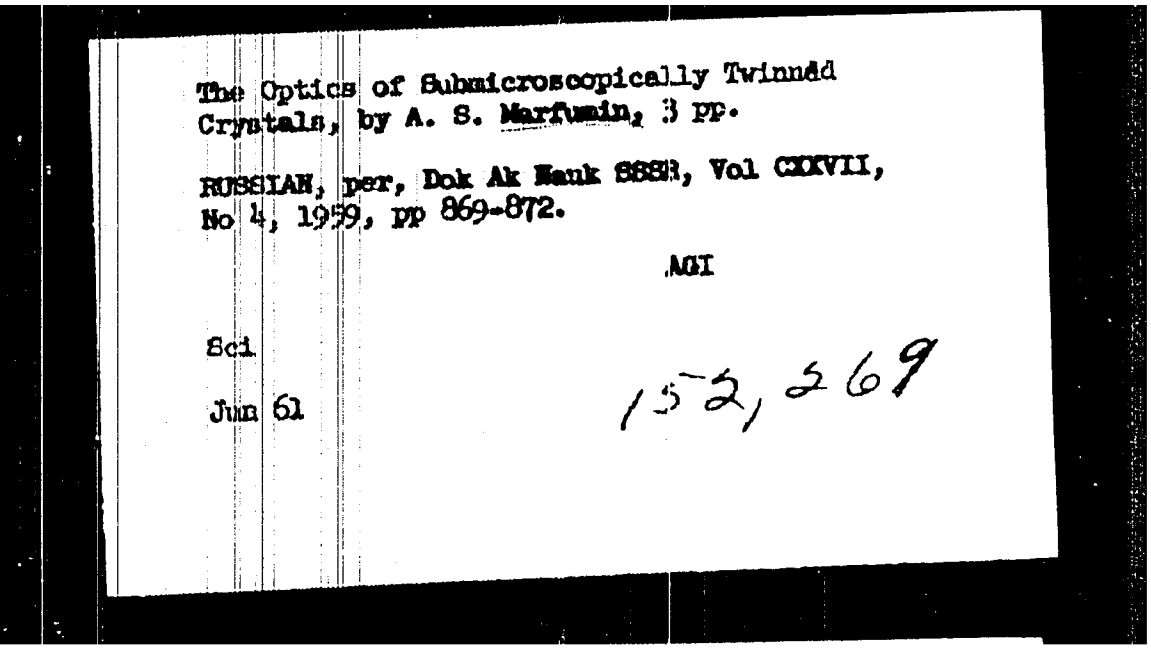

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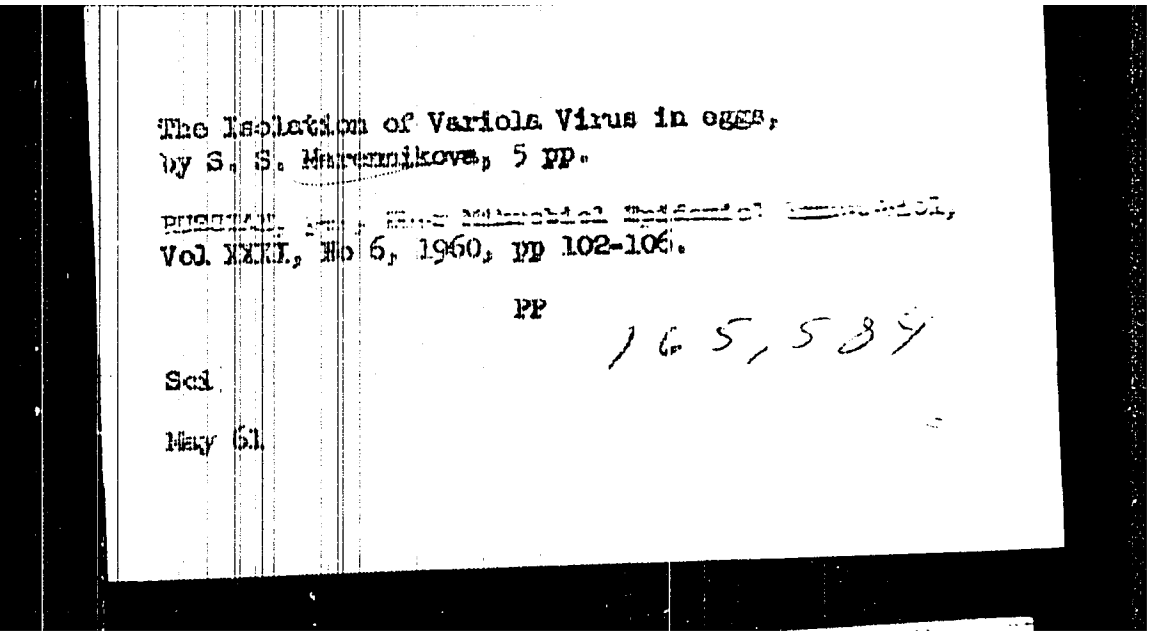

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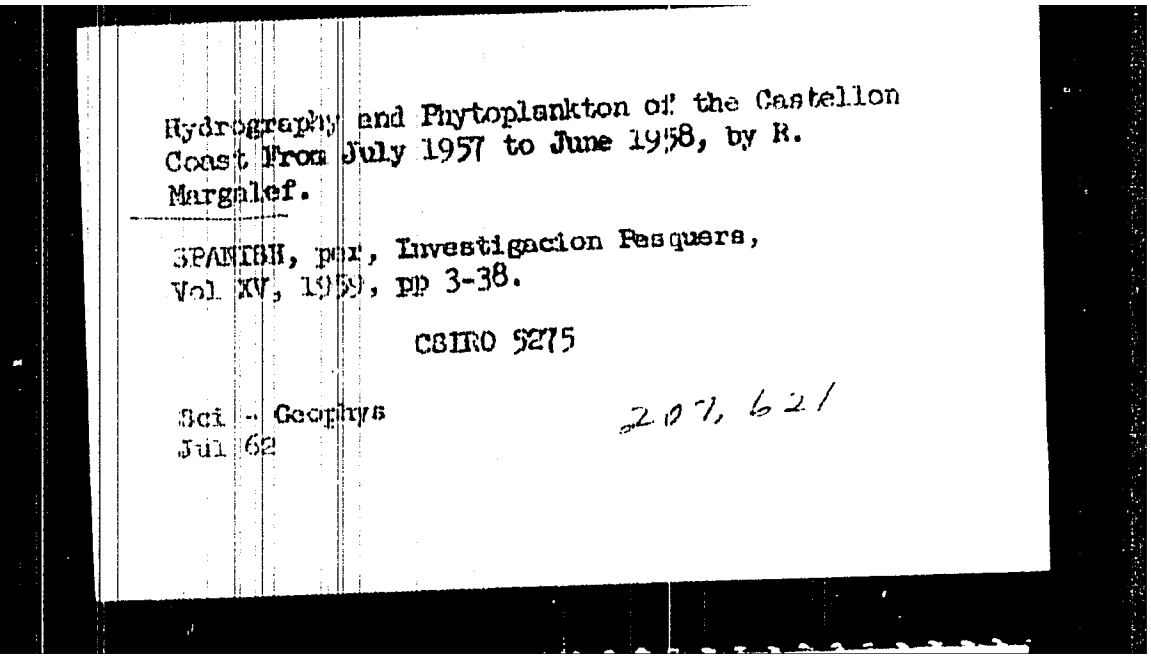

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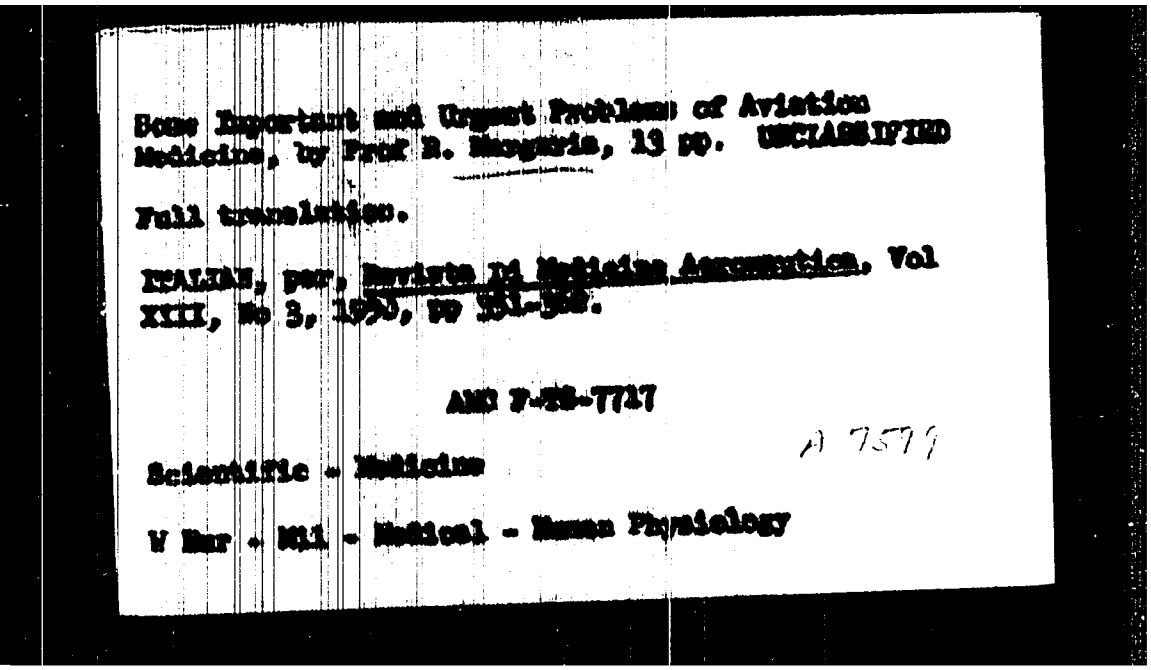

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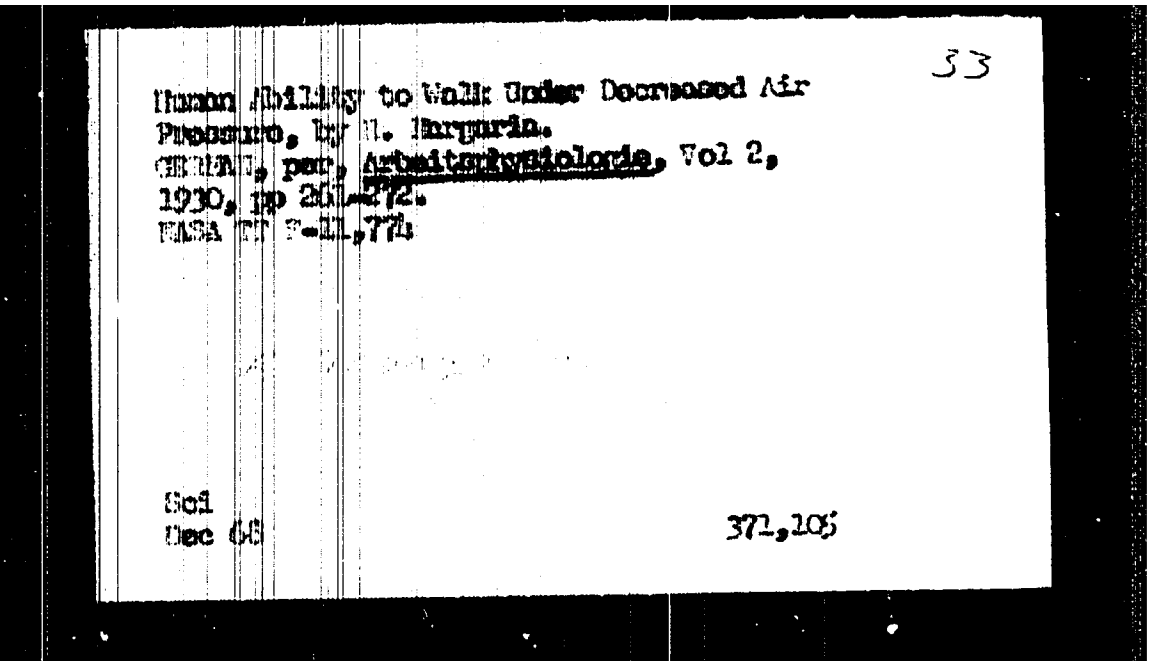

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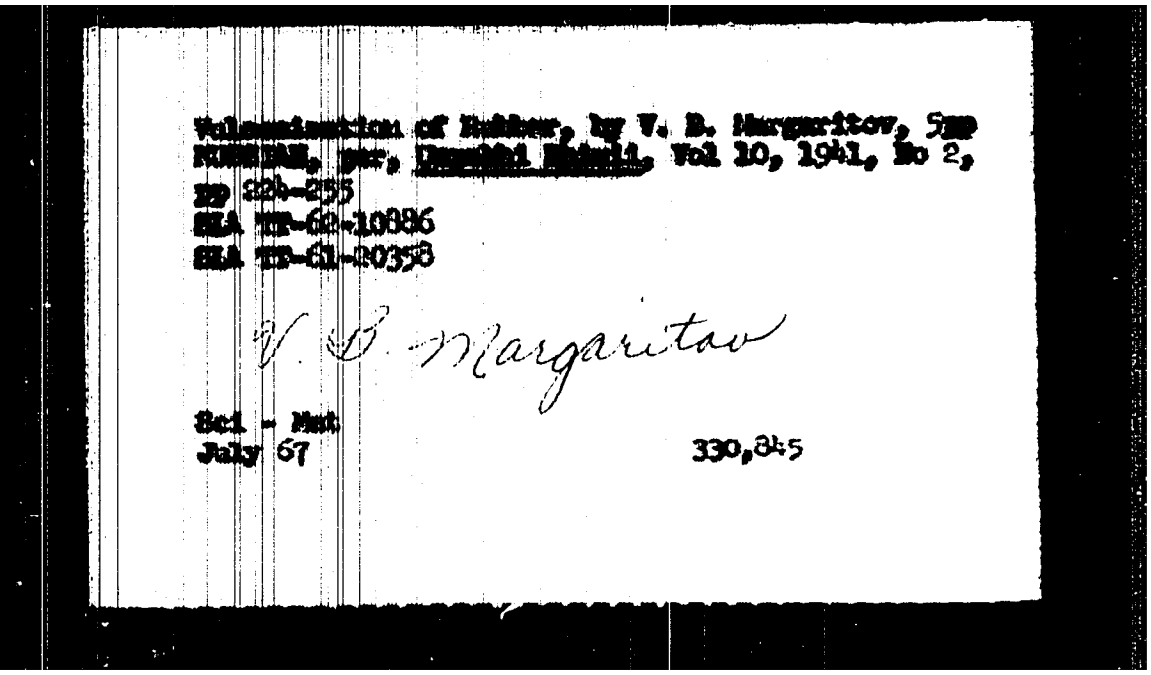

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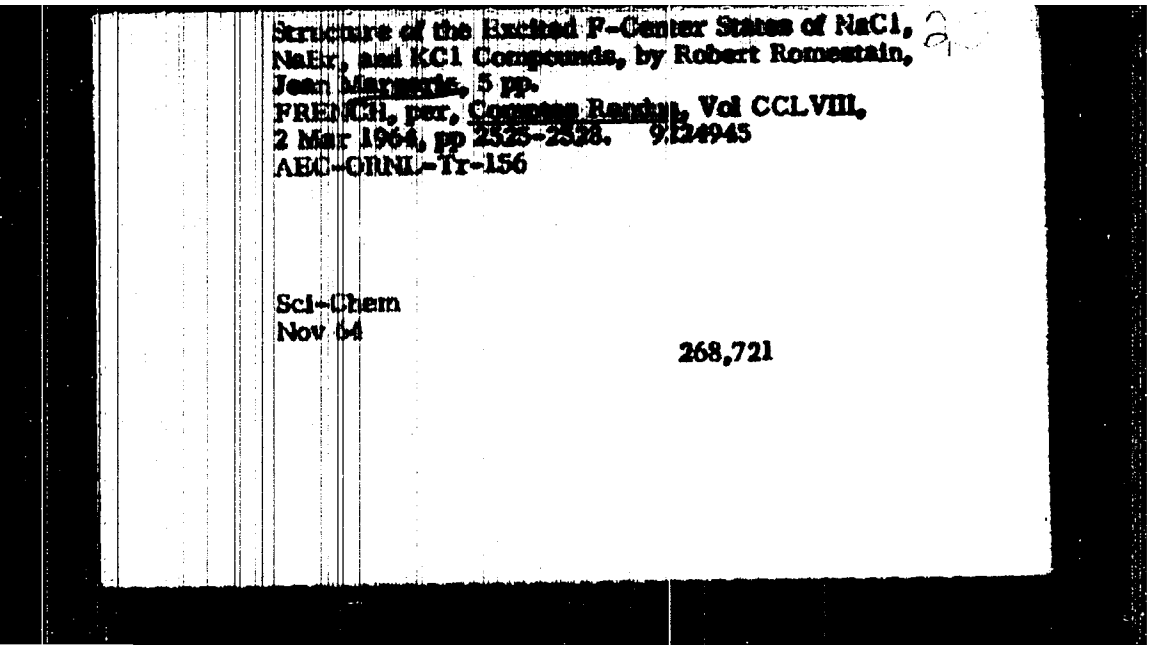

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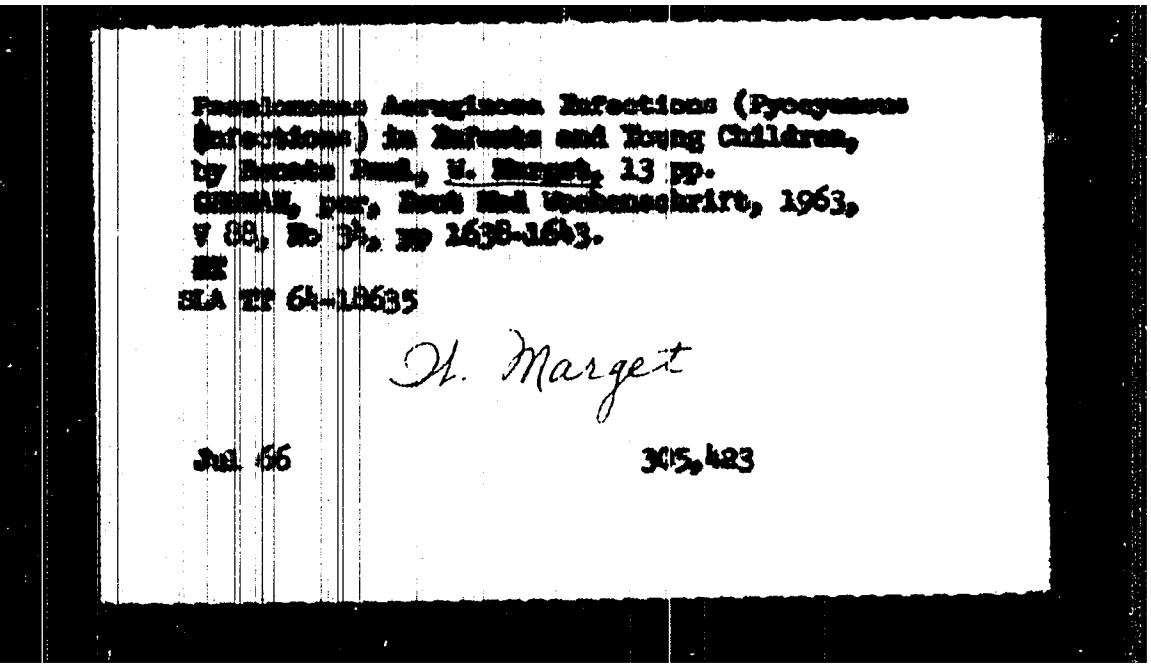

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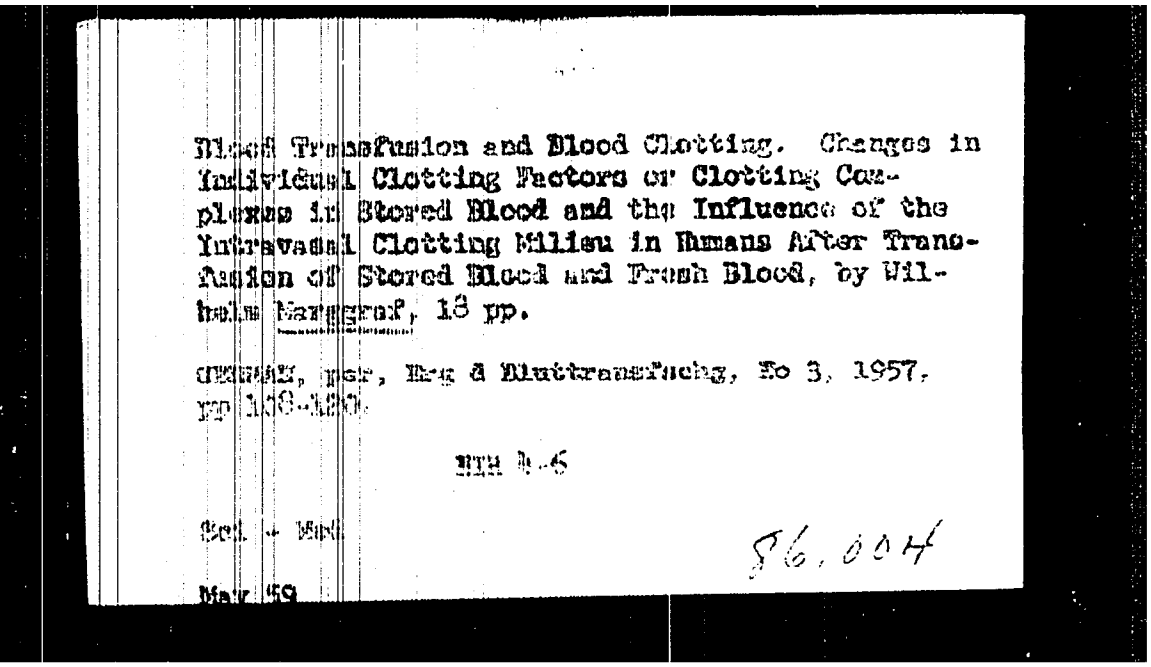

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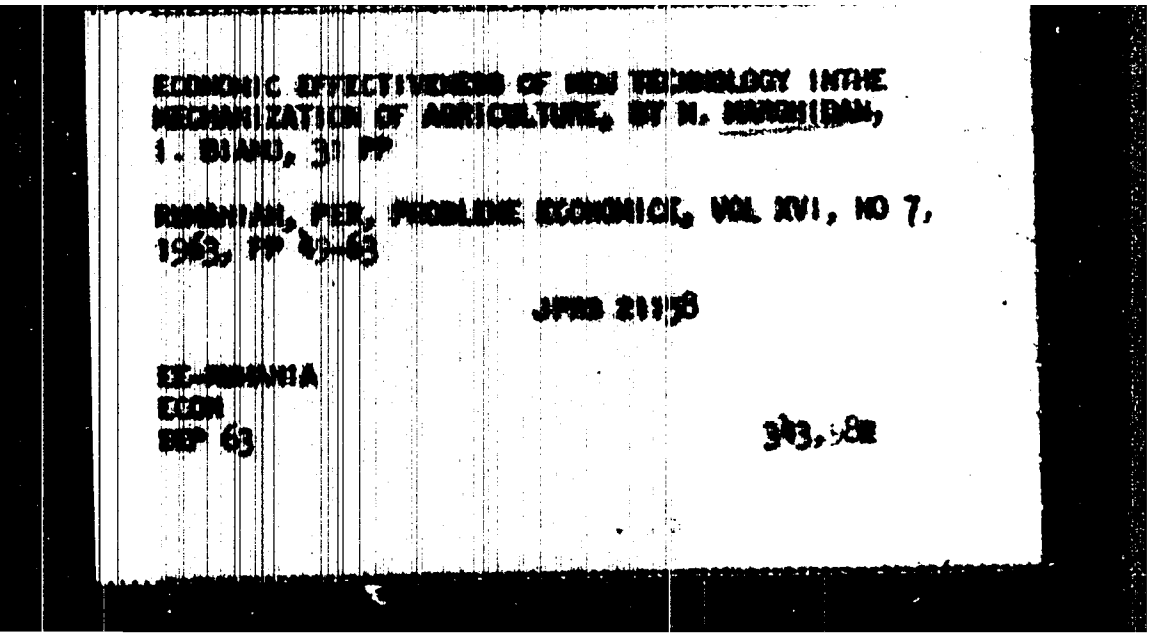

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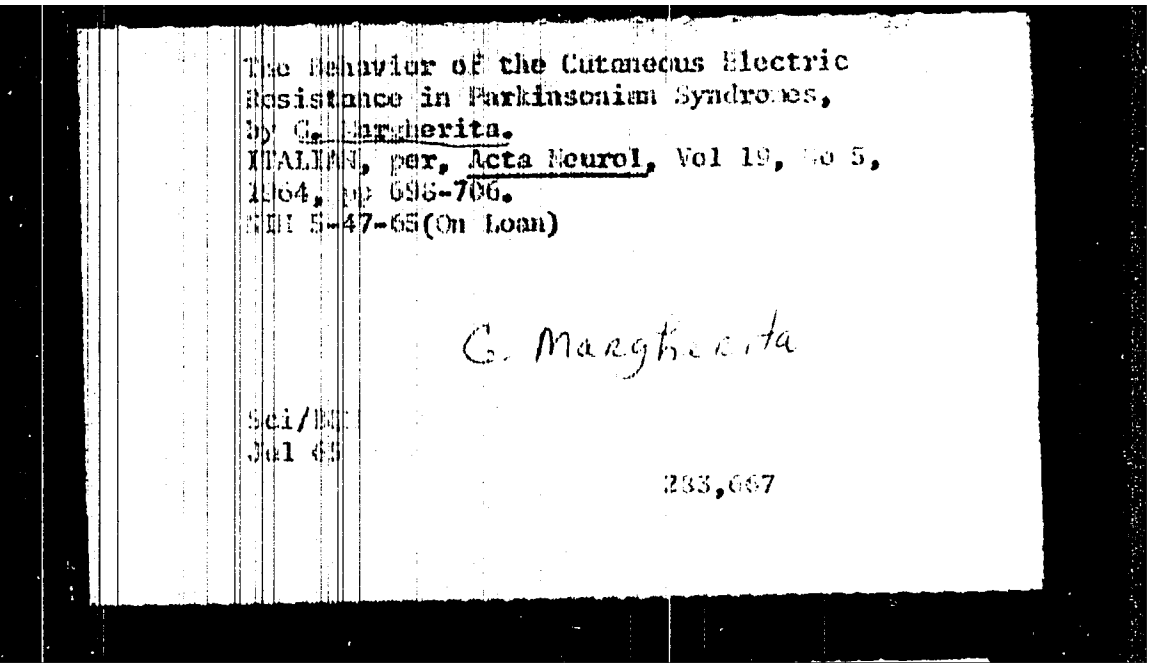

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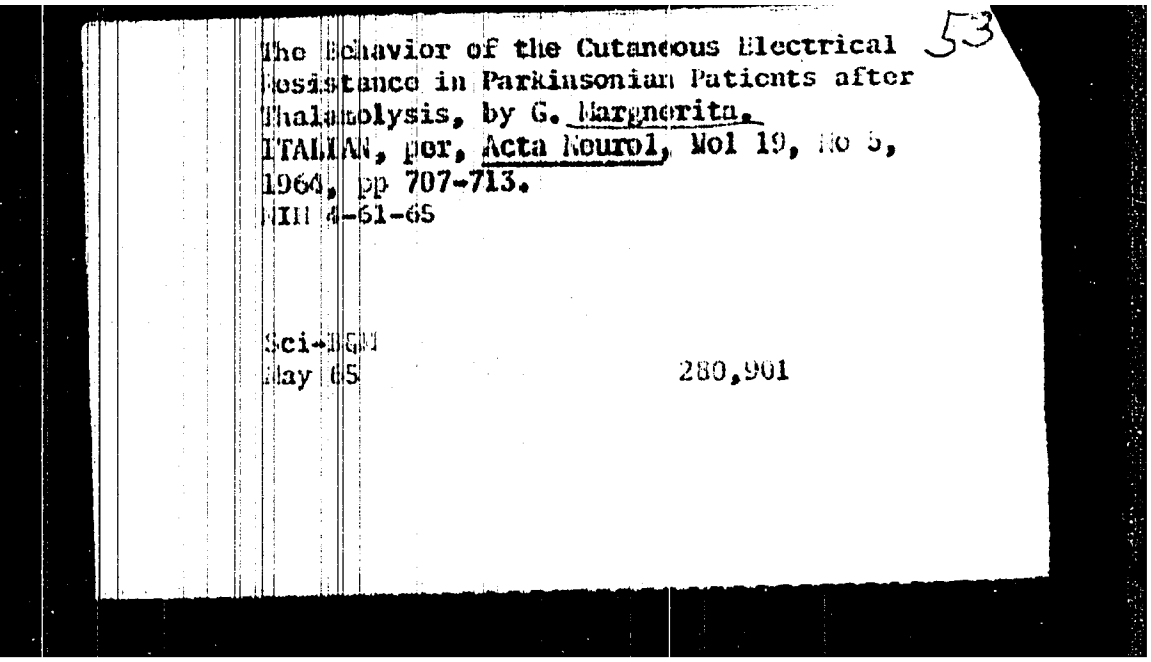

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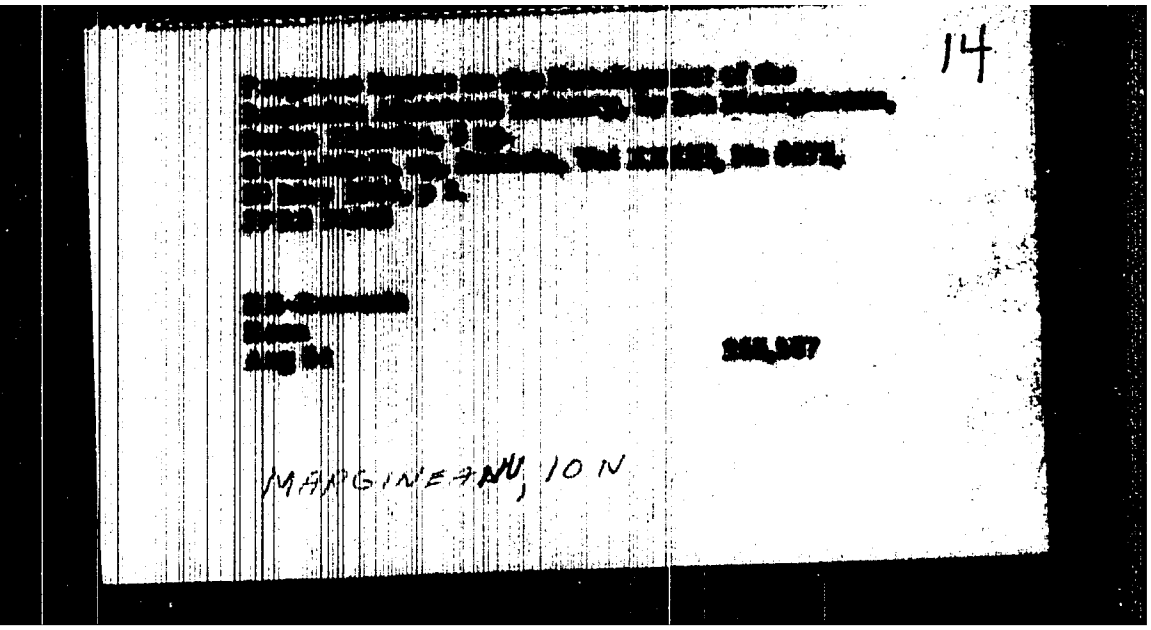

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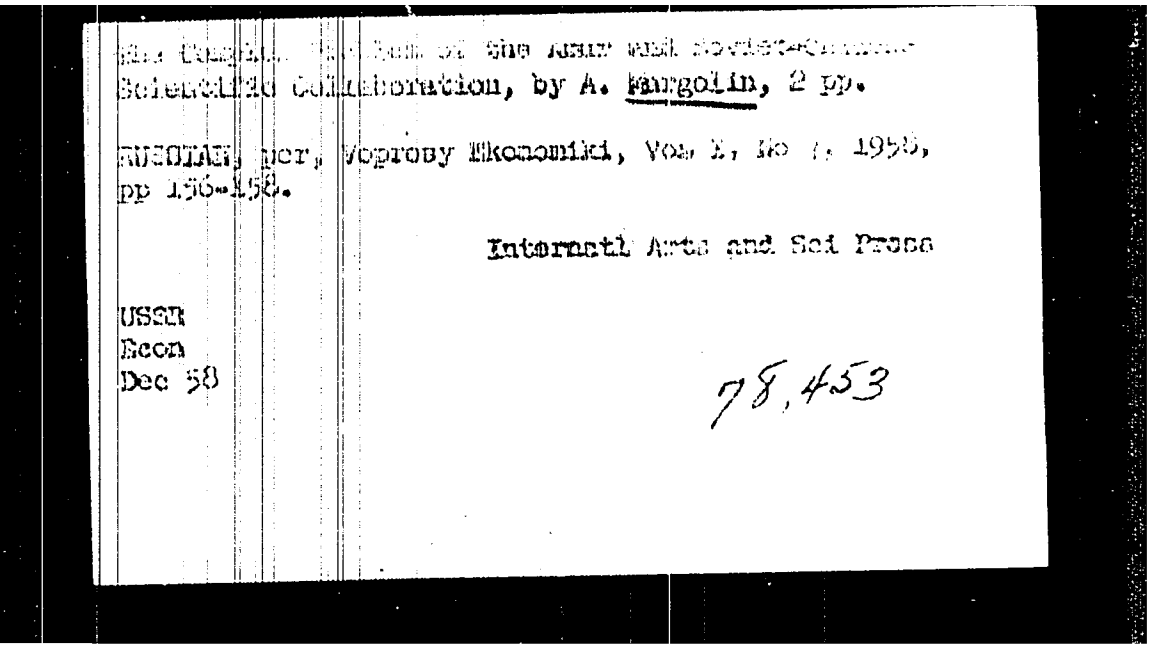

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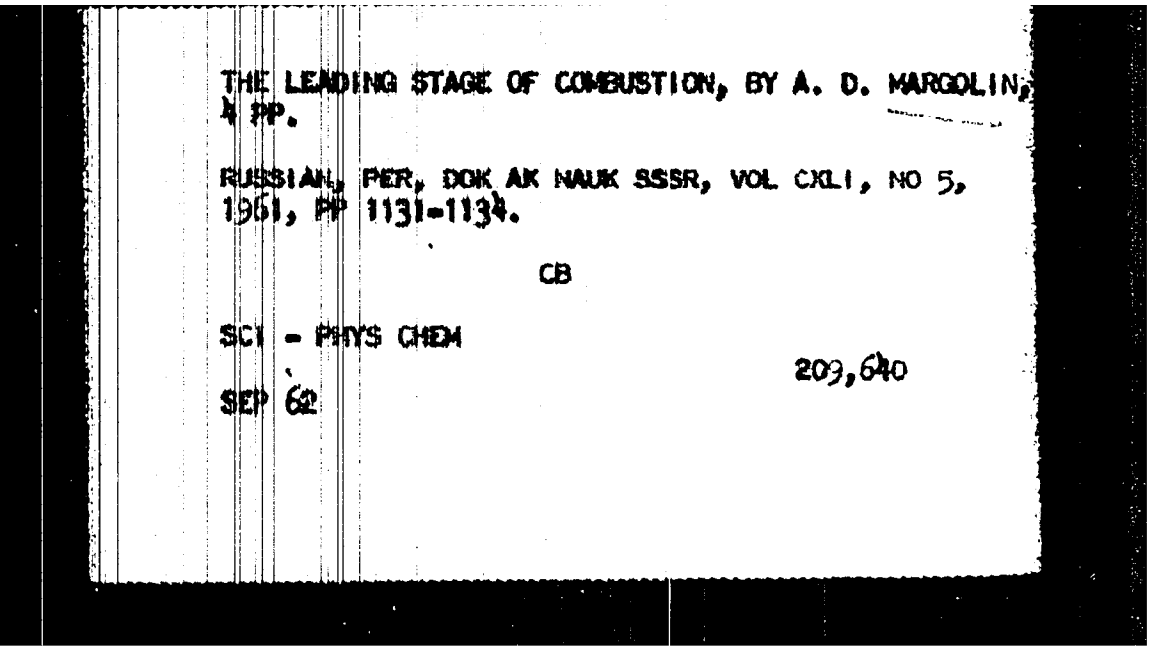

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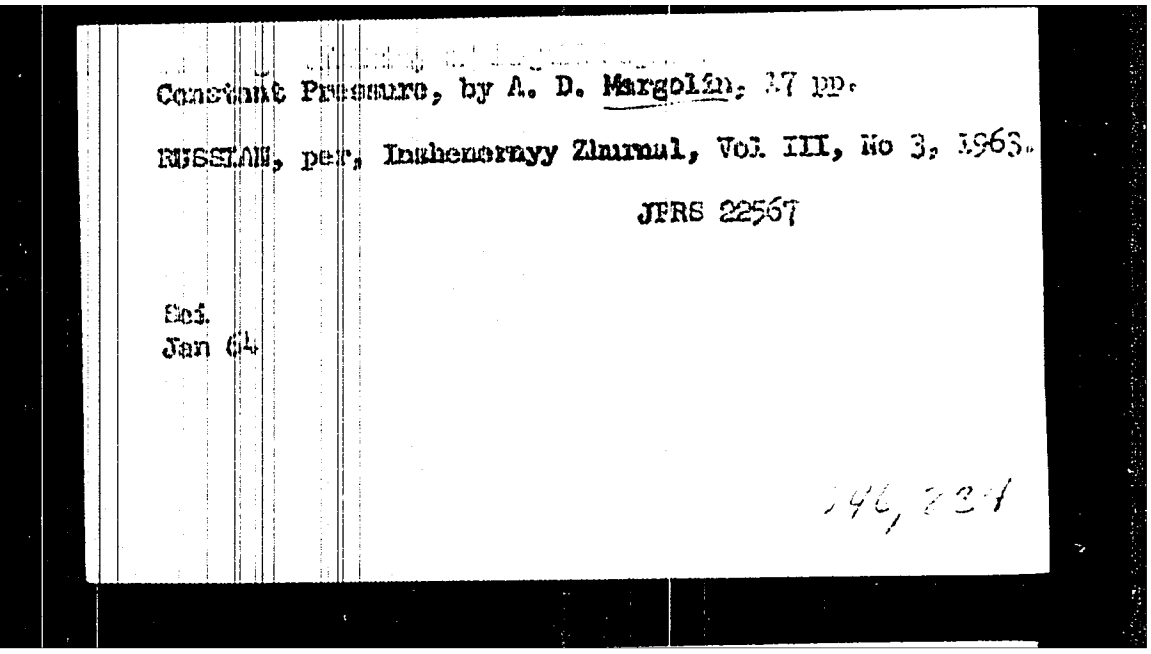

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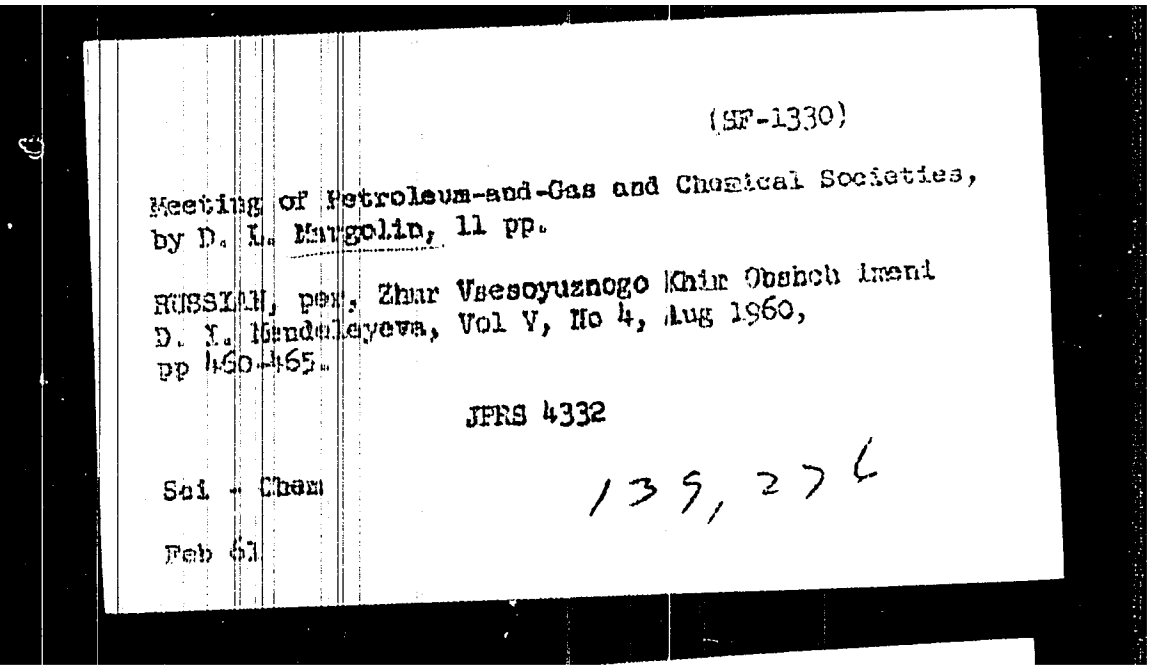

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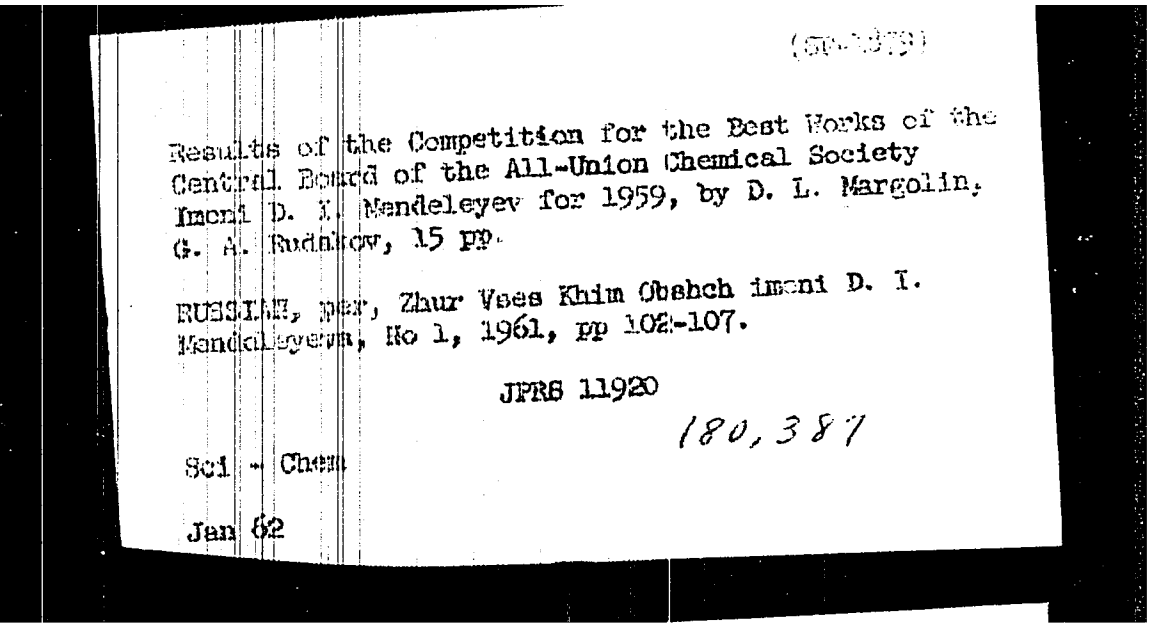

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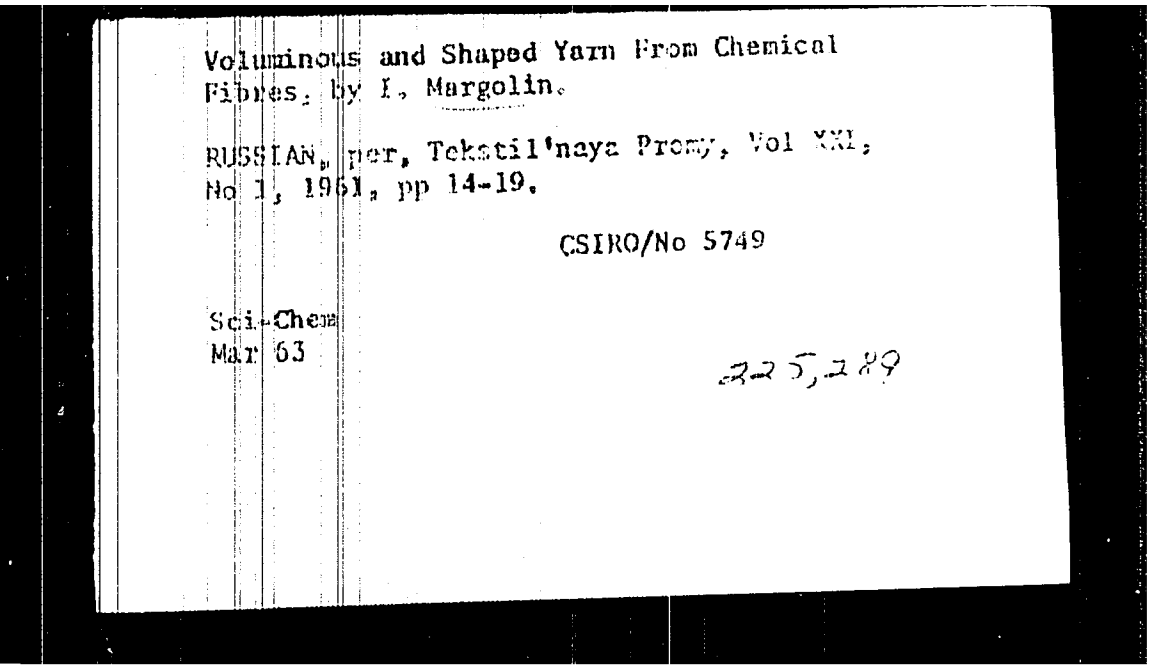

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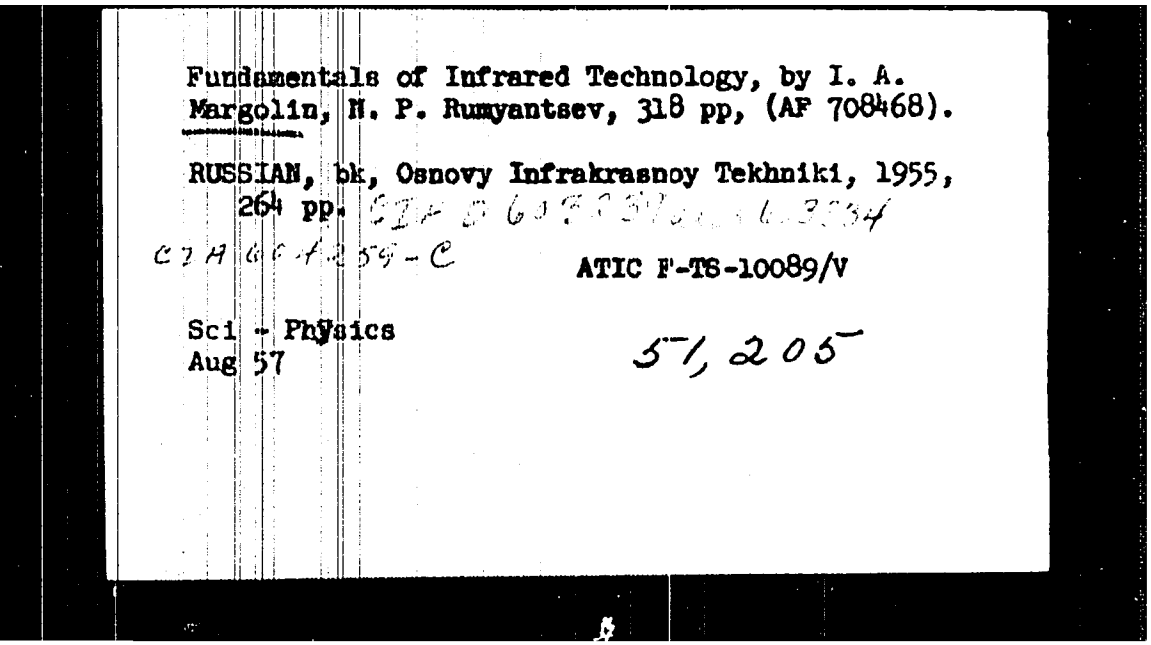

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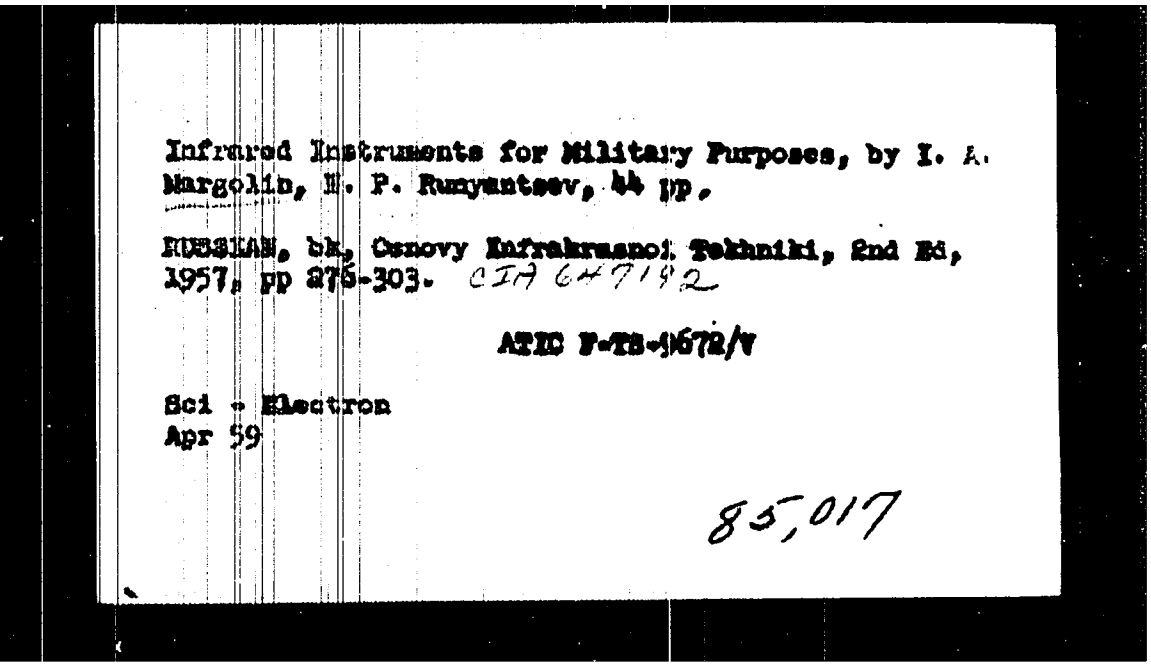

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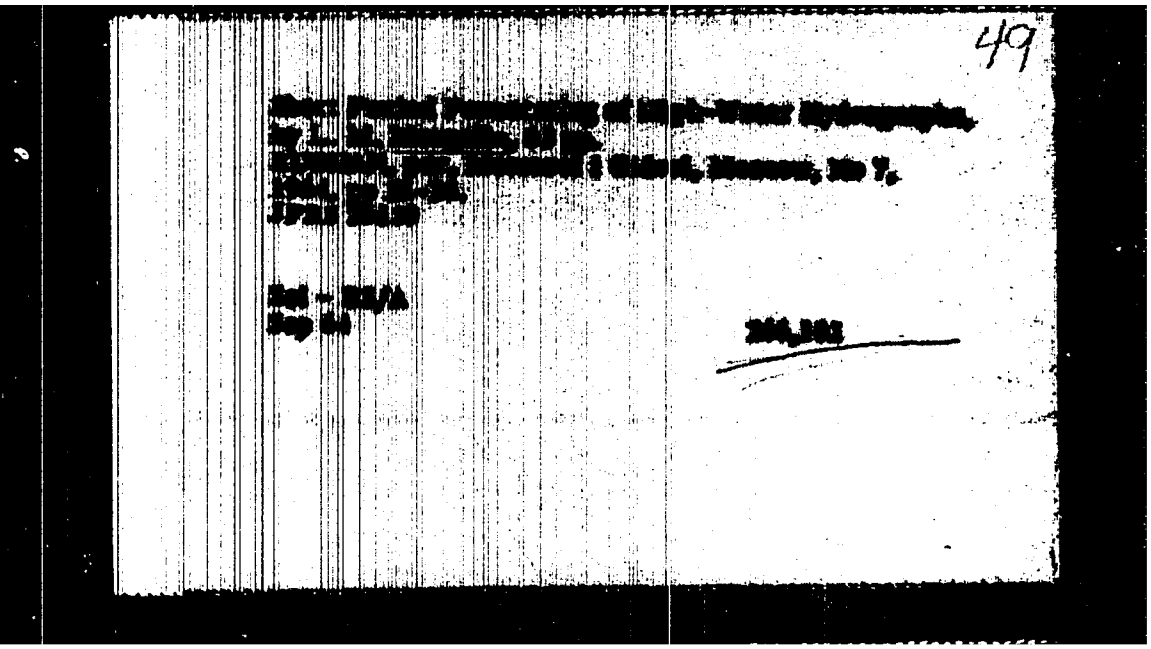

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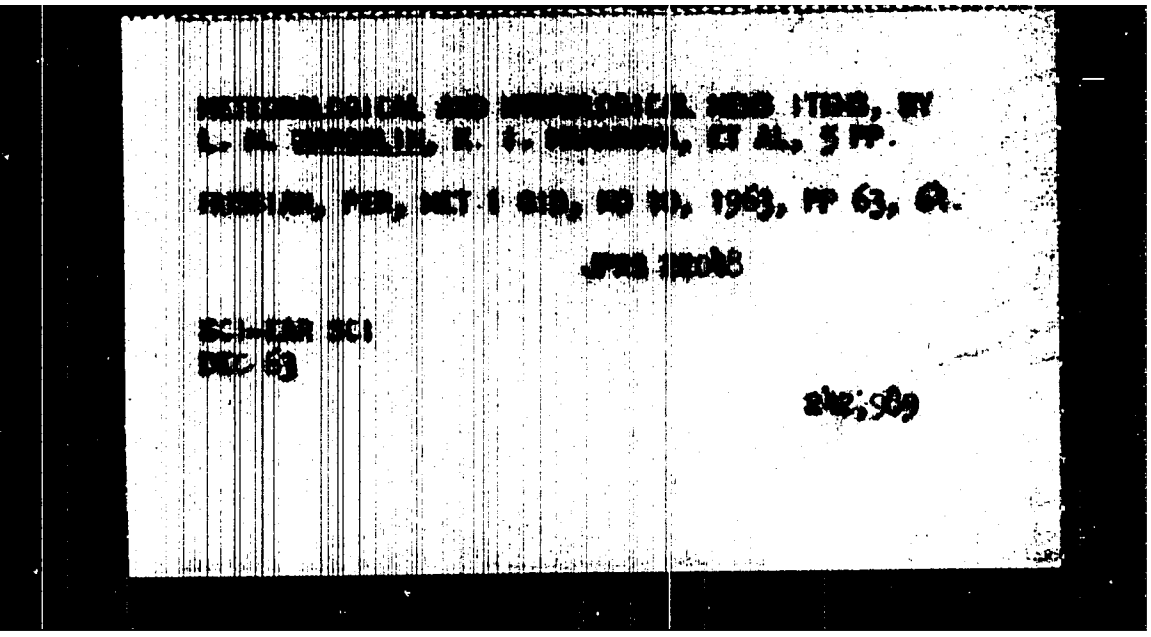

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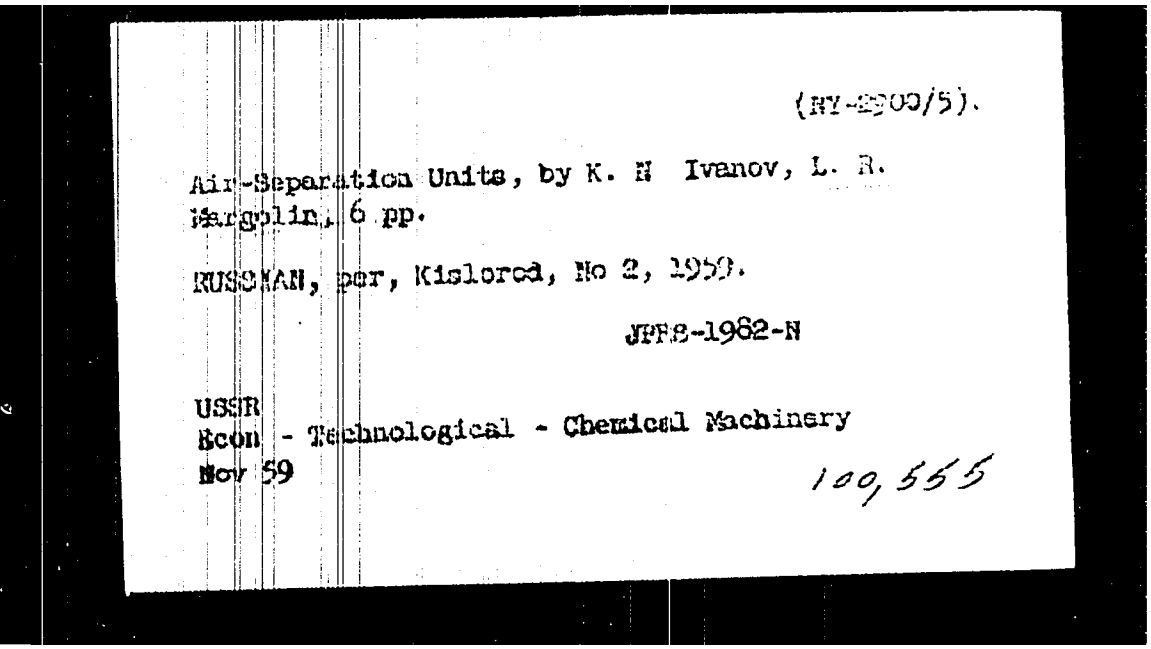

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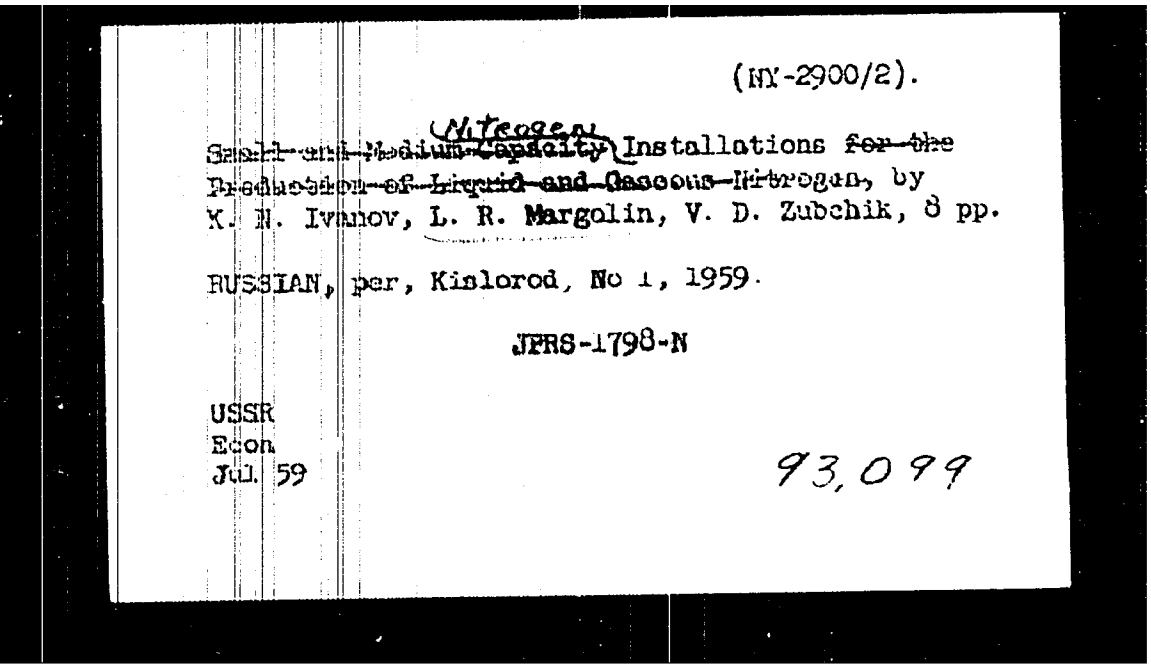

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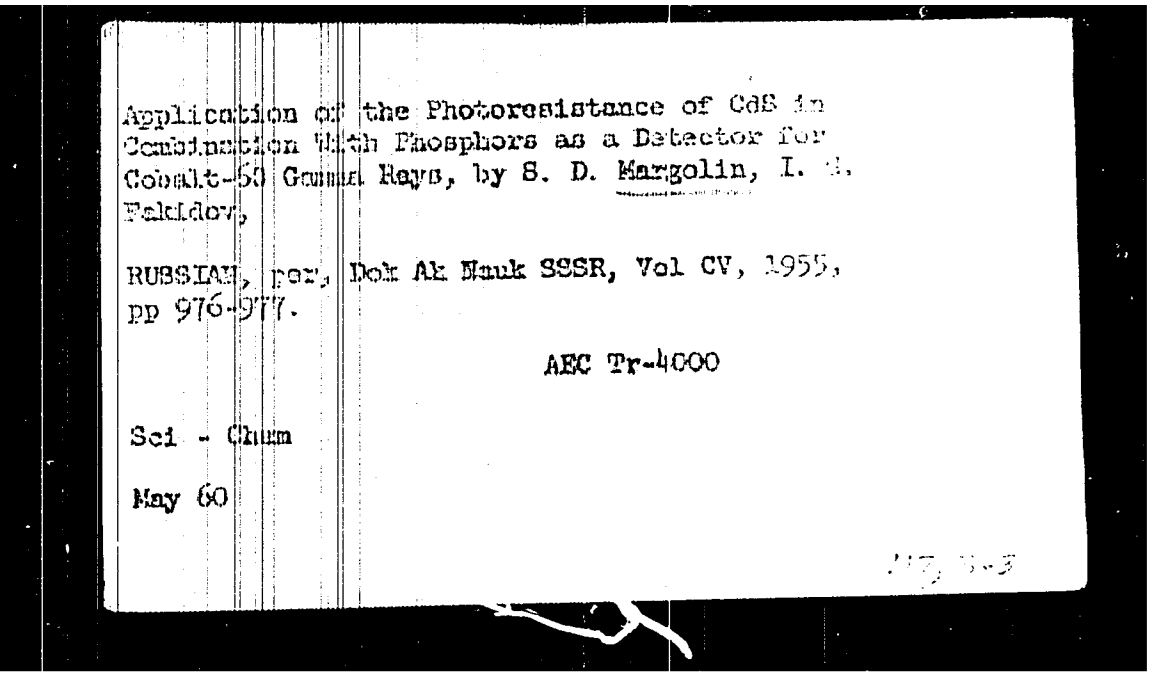

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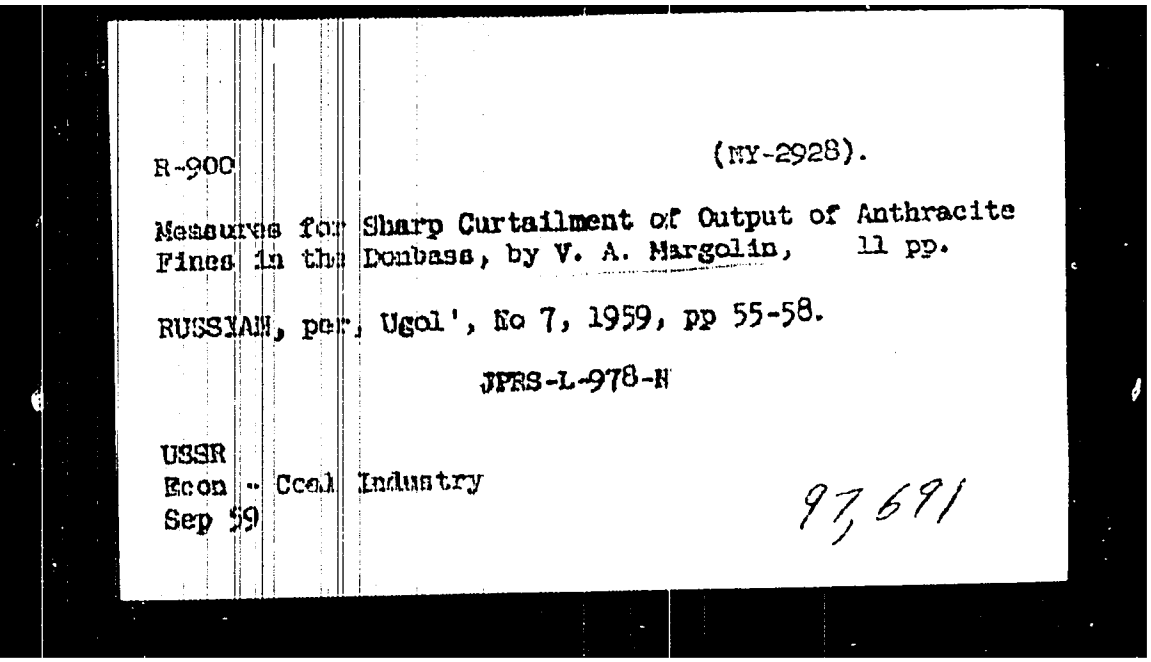

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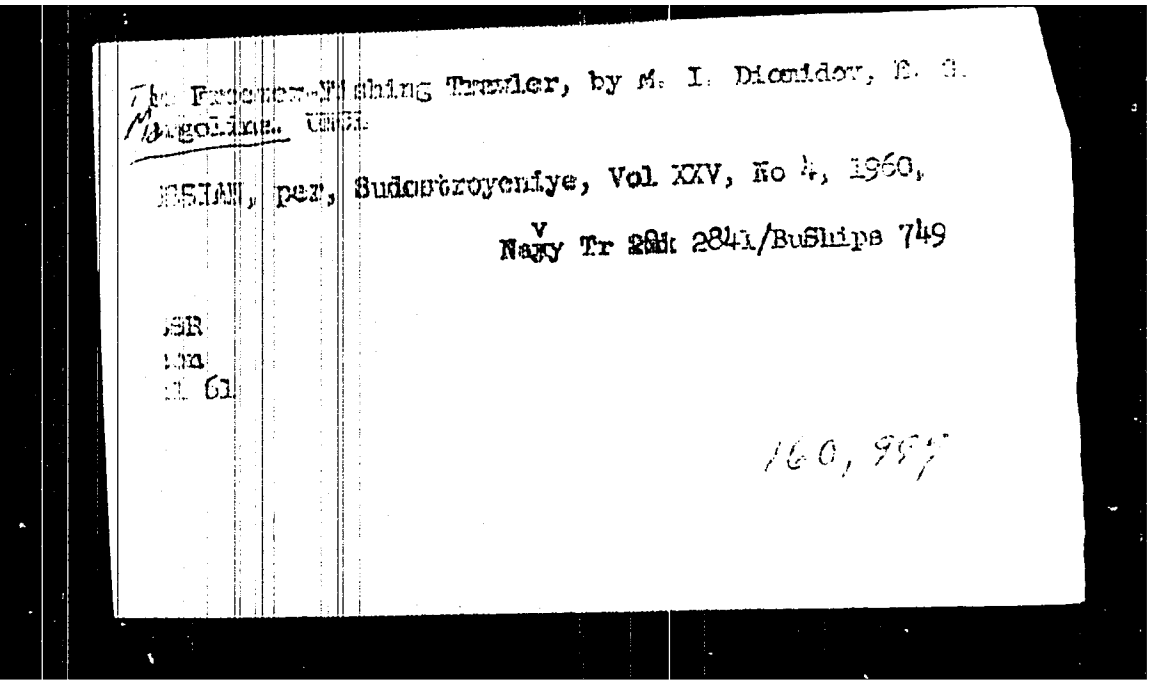

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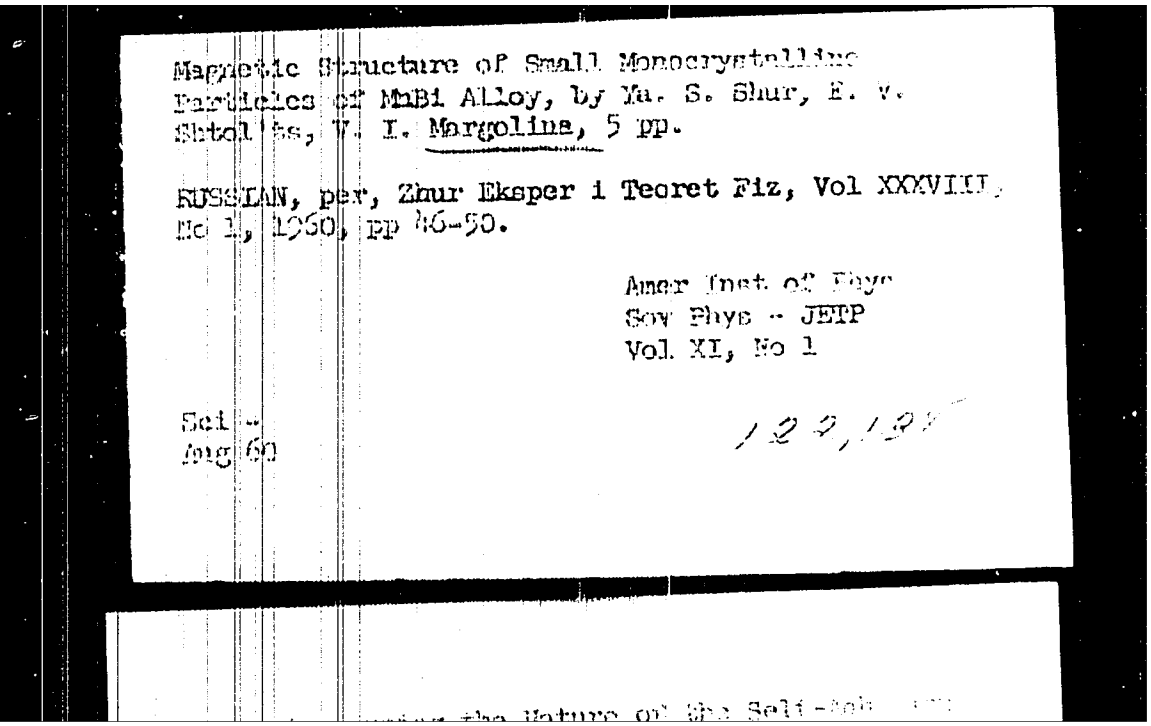

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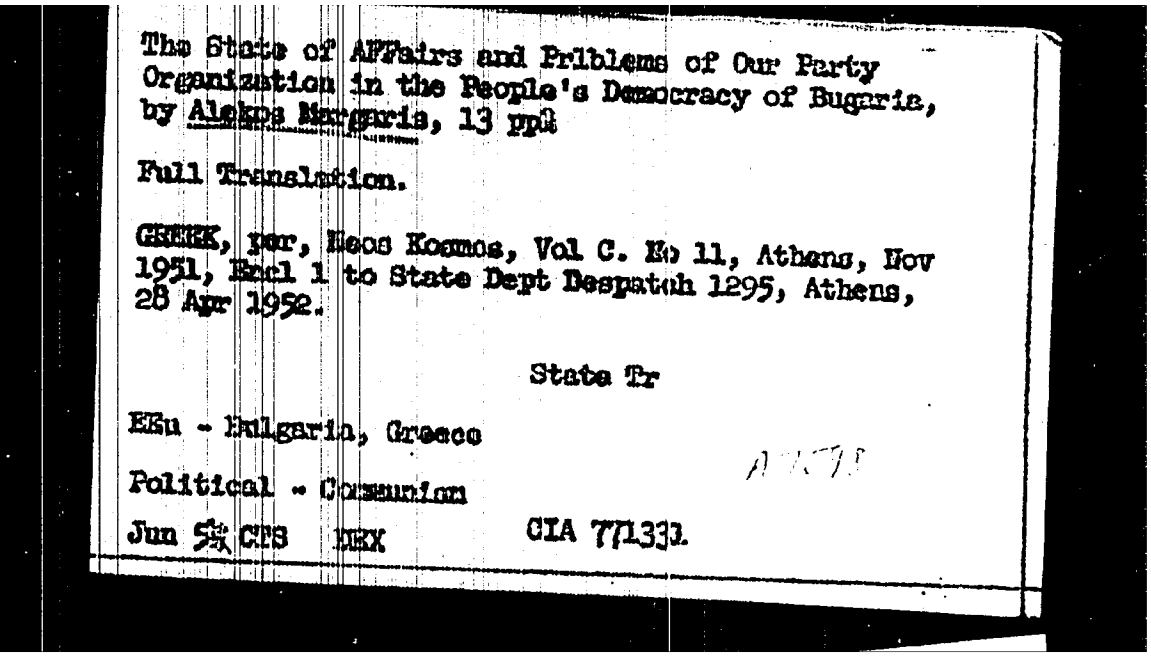

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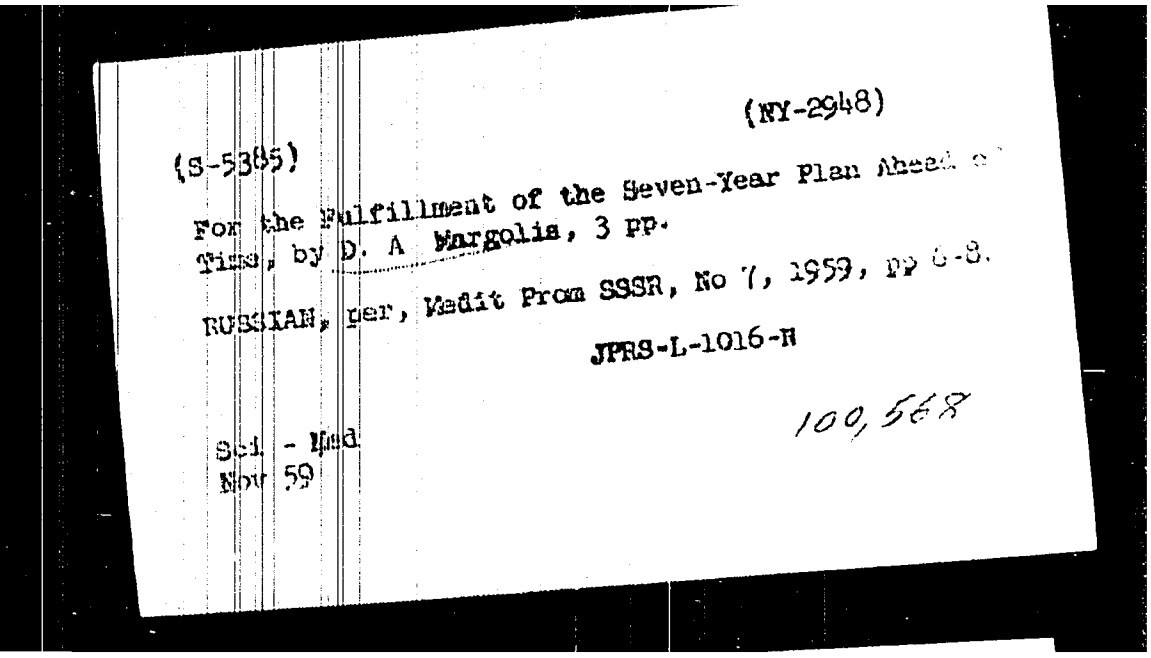

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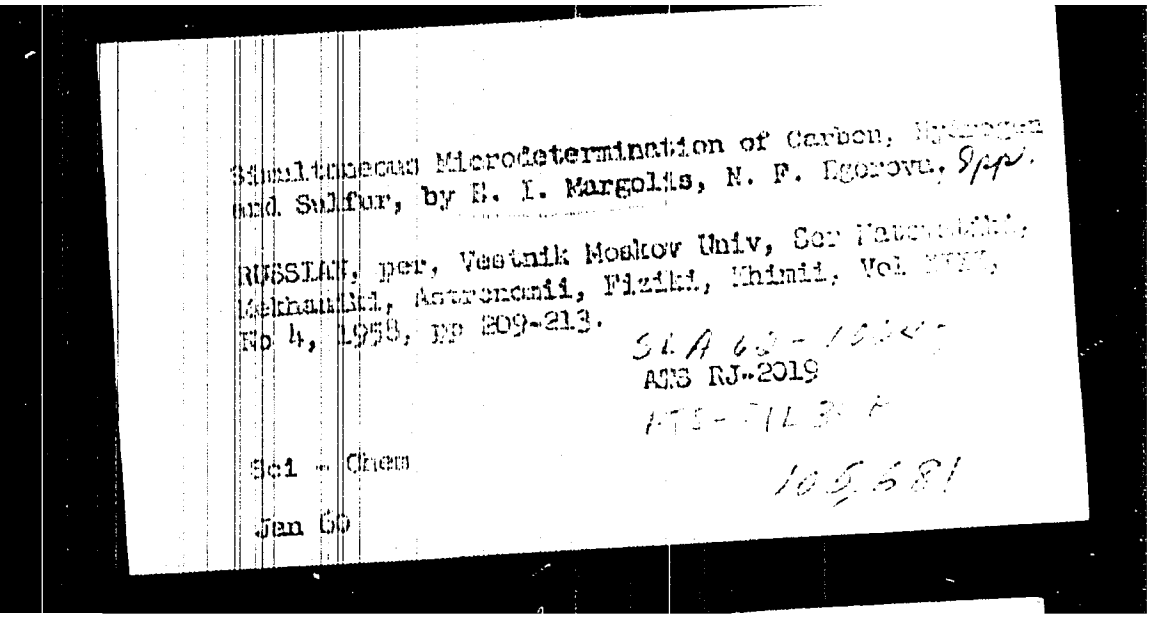

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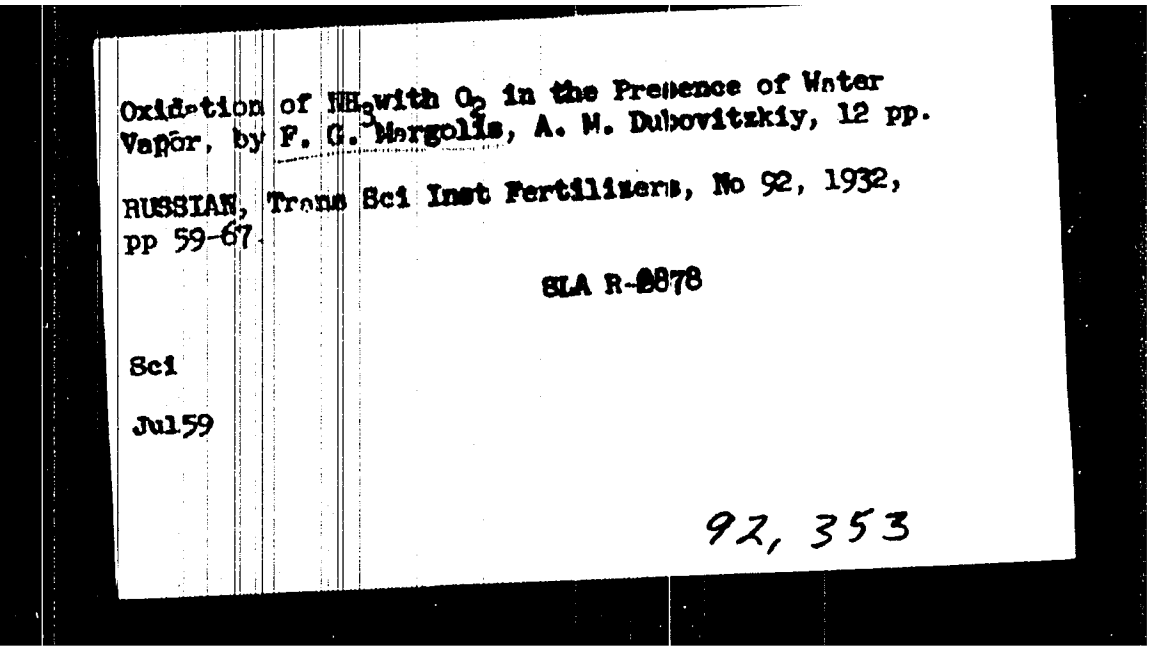

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The Dendritic Structure of the Ammonium Nitrate and its Effect on Caking of the Salt, by A. M. Dubovitskiy, F. G. Margolis, and T. V. Glazova, 6 pp.

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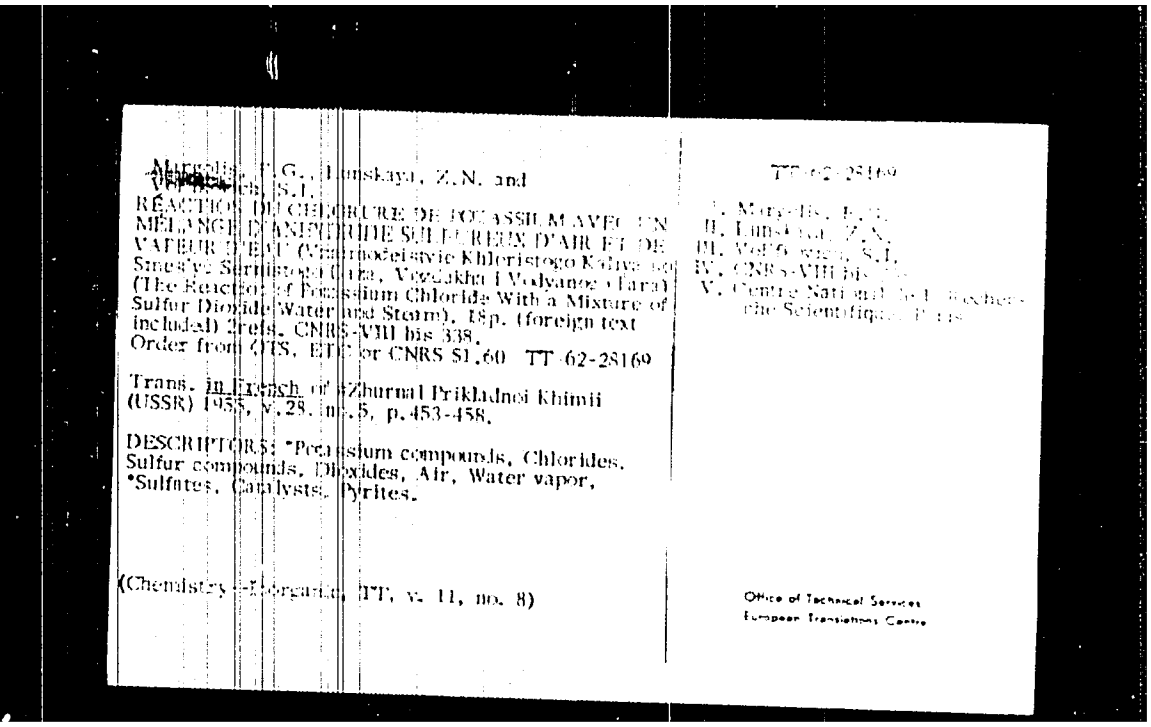

[illegible], T.G., Lunskaya, Z.N. and [illegible], S.I.
REACTION DU CHLORURE DE POTASSIUM AVEC UN MELANGE D'ANHYDRIDE SULFUREUX D'AIR ET DE VAPEUR D'EAU (Vzaimodeistvie Khloristogo Kaliya so Smes'yu Sernistogo Gaza, Vozdukha i Vodyanogo Para) (The Reaction of Potassium Chloride With a Mixture of Sulfur Dioxide, Water and Steam). 18p. (foreign text included) 2refs. CNRS-VIII bis 338.
Order from OTS, ETC for CNRS $1.60 TT-62-28169

Trans. in French of: Zhurnal Prikladnoi Khimii (USSR) 1955, v.28, no.5, p.453-458.

DESCRIPTORS: *Potassium compounds, Chlorides, Sulfur compounds, Dioxides, Air, Water vapor, *Sulfates, Catalysts, Pyrites.

(Chemistry - Inorganic, TT, v. 11, no. 8)

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I. [illegible], T.G.
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The State and Trends of Fertilizer Manufacture by the Nitric Acid Treatment of Phosphates in the Capitalist Countries, by A. M. Dubovitsky, F. O. Margolis, 10 pp.

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RUSSIAN, no per, Zhur Prik Khim, Vol XXIX, No 3, 1956, pp 321-329. CIA 9000995-C, Reel No 42

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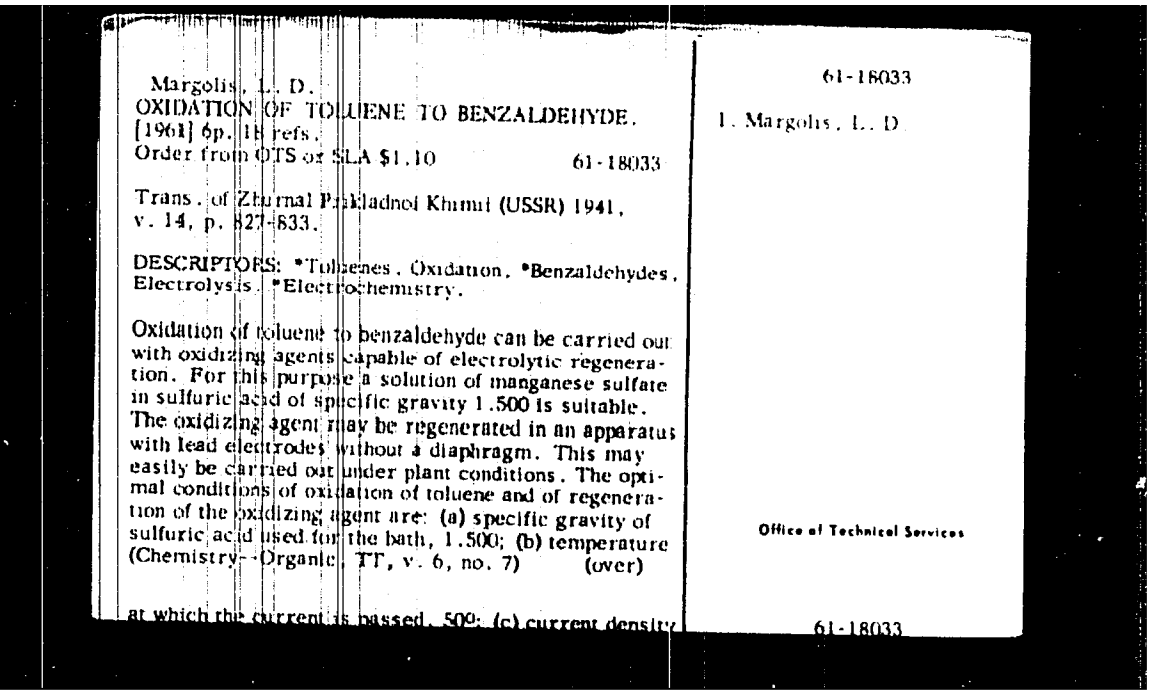

Margolis, L. D.
OXIDATION OF TOLUENE TO BENZALDEHYDE.
[1961] 6p. 18 refs.
Order from OTS or SLA $1.10 61-18033

Trans. of Zhurnal Prikladnoi Khimii (USSR) 1941, v. 14, p. 827-833.

DESCRIPTORS: *Toluenes, Oxidation, *Benzaldehydes, Electrolysis, *Electrochemistry.

Oxidation of toluene to benzaldehyde can be carried out with oxidizing agents capable of electrolytic regeneration. For this purpose a solution of manganese sulfate in sulfuric acid of specific gravity 1.500 is suitable. The oxidizing agent may be regenerated in an apparatus with lead electrodes without a diaphragm. This may easily be carried out under plant conditions. The optimal conditions of oxidation of toluene and of regeneration of the oxidizing agent are: (a) specific gravity of sulfuric acid used for the bath, 1.500; (b) temperature (Chemistry--Organic, TT, v. 6, no. 7) (over)

at which the current is passed, 50°; (c) current density

61-18033

L. Margolis, L. D

Office of Technical Services

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Modification of Catalysts for the Oxidation of Hydrocarbons, by L. Ya. Margolis, E. Kh. Enikeyev, et al., 7 pp.

RUSSIAN, per, Kinetika i Kataliz, Vol III, No 2, 1962, pp 181-188.

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Kinetics of the Oxidation of Propylene to Acrolein on Copper Catalyst, by O. V. Isayev, L. Ya. Margolis, 4 pp.

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The Mechanism of the Oxidation of Propylene to Acrolein on a Cuprous Oxide Catalyst, by O. V. Isayev, L. Ya. Margolis, I. S. Sazonov, 4 pp.

RUSSIAN, per, Dok Ak Nauk SSSR, Vol CXXIX, No 1, 1959, pp 141-144.
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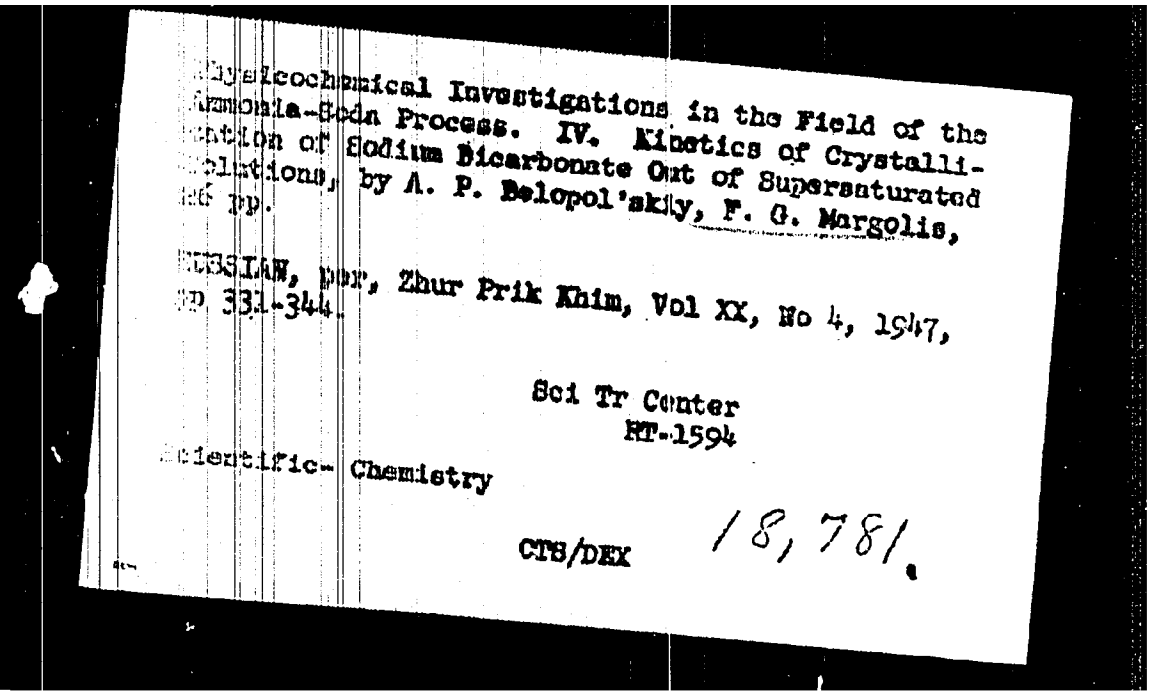

Physicochemical Investigations in the Field of the Ammonia-Soda Process. IV. Kinetics of Crystallization of Sodium Bicarbonate Out of Supersaturated Solutions, by A. P. Belopol'skiy, F. G. Margolis,

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Fundamental Relationships in the Catalytic Oxidation of Hydrocarbons. I. Adsorption of Hydrocarbons on Oxidation Catalysts, by L. Ya. Margolis.

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A Study of Oxygen Exchange on Vanadium Catalysts by the Use of O^{18}, by L. Ya. Margolis and E. G. Plyshevskaya, 5 pp.

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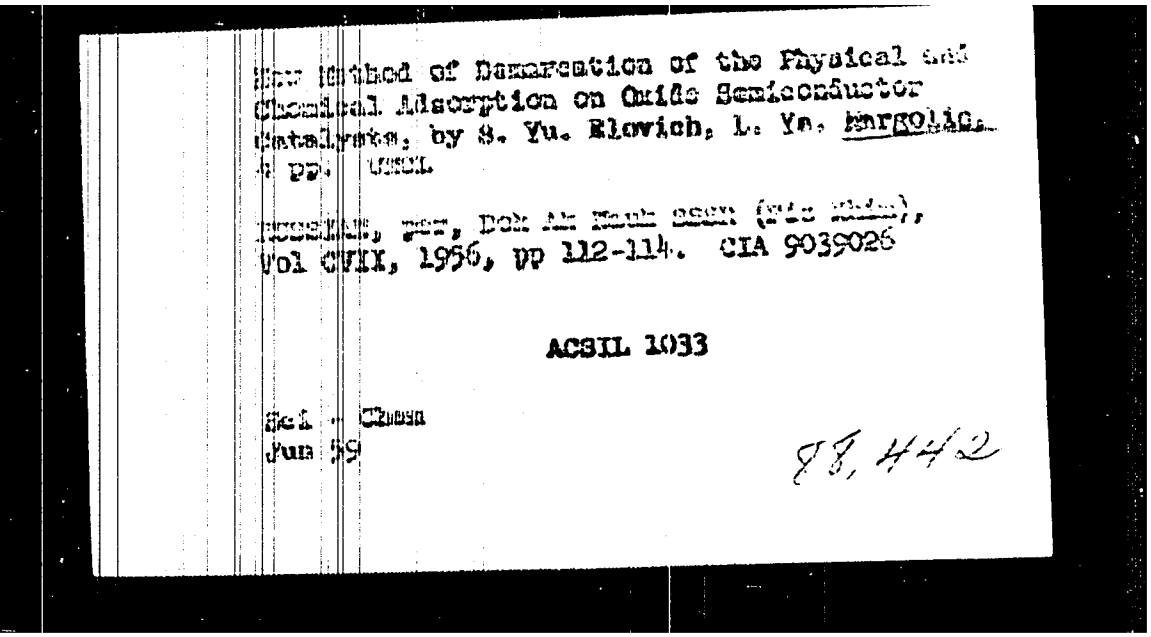

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Catalytic Oxidation of Propylene to Acrolein. I,
by O. V. Isayev, I. Ya. Margolis, S. Z. Roginskiy,
5 pp.

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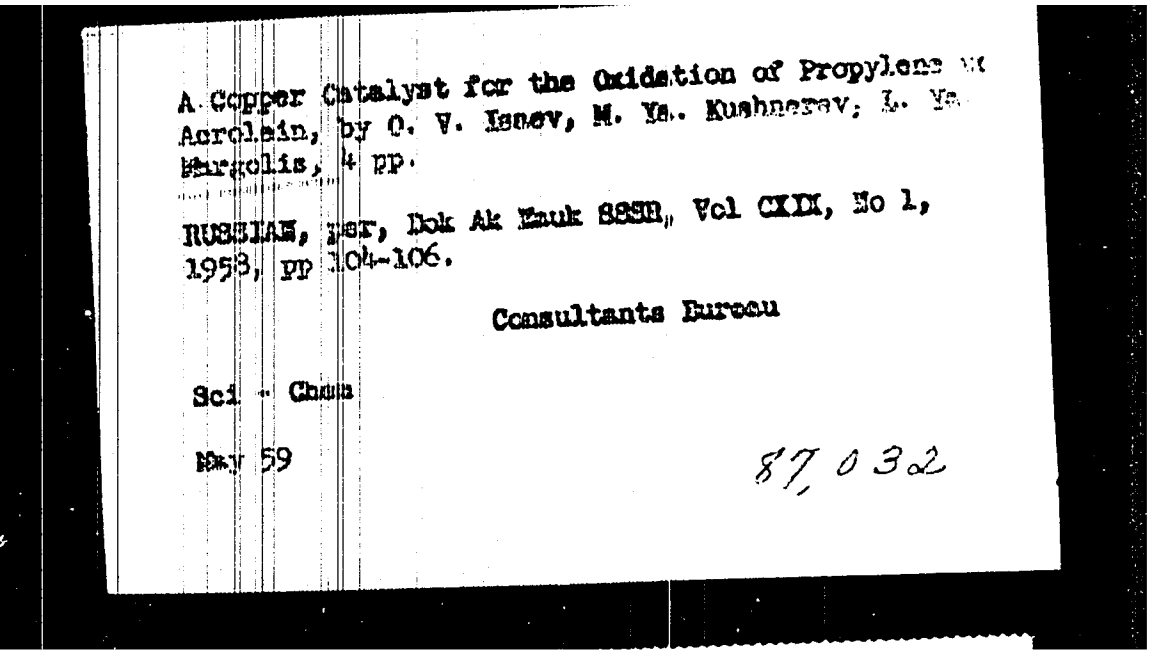

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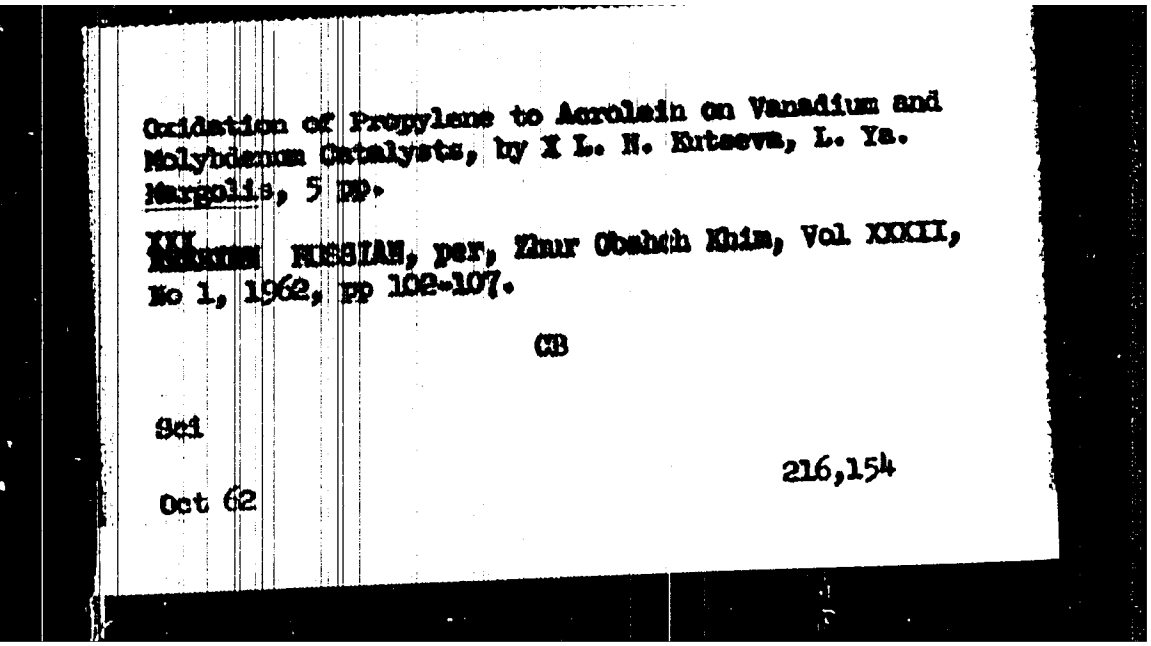

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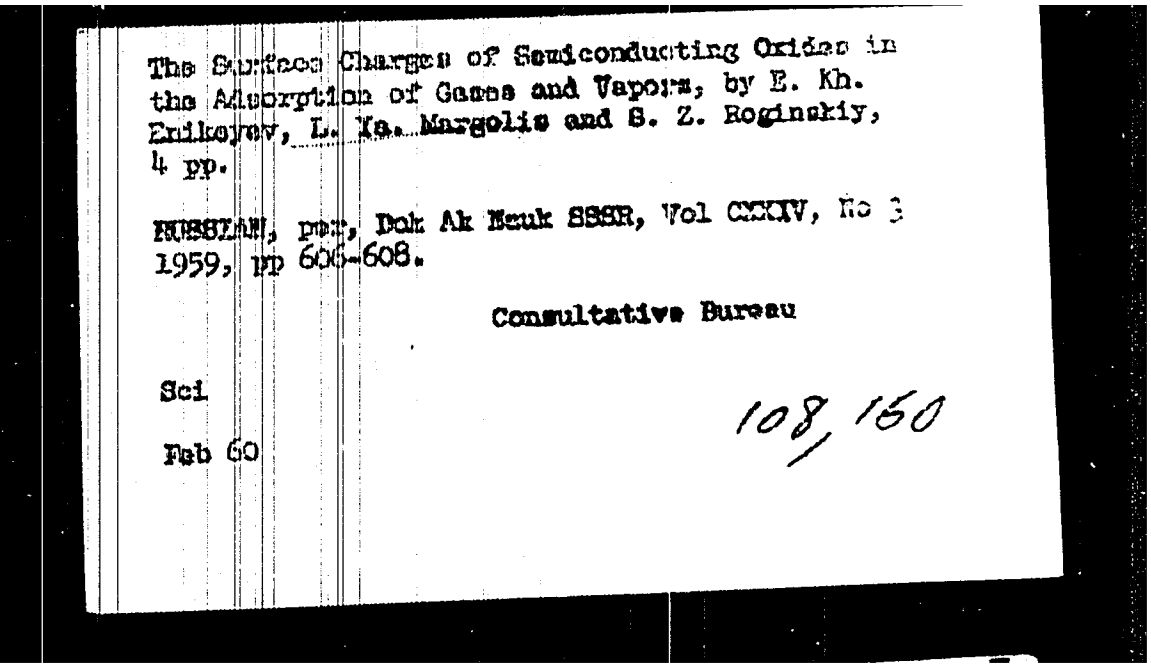

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